### https://doi.org/10.48047/AFJBS.6.7.2024.957-985



# THE NEW INSIGHT OF NANOSPONGES IN TOPICAL DRUG DELIVERY Poonam.P. Patil<sup>1</sup>\*, Dr. Sunil J. Aher<sup>2</sup>, Dr. Indrayani D. Raut<sup>3</sup>, Dr. Sandip Bandgar<sup>4</sup>, Pallavi S. Ghule<sup>2</sup>

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#### ABSTRACT

Skin-related diseases are a leading cause of illness and require precise treatment without harming the body. There are a variety of treatments available that have various side effects. Nanotechnology offers advantages like higher therapeutic efficiency, property of drug targeting, and reduction in toxicity. Nanosponges involve porous, spongelike surfaces, and colloidal sizes less than 1 µm in diameter which allows for high loading capacity. Nanosponges are simple to alter the pharmaceutical release contour and enhance formulation stability while reducing the drug's adverse effects. The ability of Nanosponges to selfsterilize is their most notable feature. They are used as antiallergic, non-mutagenic, and non-irritant in many investigations. In this review, the collection of NSs, with and characterization preparation parameters and applications in various fields is included. This review will be quite useful in the future in exploring the Nanosponge delivery system in different areas.

**Keywords:** Nanosponge, cancer therapy, cyclodextrin, solvent methods, optimization of nanosponges.

Volume 6, Issue 7, June 2024

Received: 25 April 2024

Accepted: 03 june 2024

Published: 20 June 2024 doi:

10.48047/AFJBS.6.7.2024.957-985

### 1. Background

Skin covers approximately 2 m<sup>2</sup> of the human body.<sup>1</sup> Skin-related diseases are a third-leading cause of illness and one of the top ten causes of disability.<sup>2</sup> Skin-related diseases involve Acne, Atopic Dermatitis, Psoriasis, Impetigo, Scleroderma, and Vitiligo caused by fungal or bacterial infections, allergens, or genetic factors. Antibiotics, medicated creams, ointments, or gels, laser treatment, steroid pills, and surgical procedures are among the treatment options available. However, they have numerous drawbacks that render them ineffective in treating skin infections. Local anaesthesia was required for the majority of skin surgery.<sup>3</sup> Vehicles, creams, and solutions have drawbacks because the of quick evaporation of water/alcohol, that is measured as cooling, affects spreading on skin, resulting in an inconsistent topical dosage form.<sup>4</sup> Steroids may lead to melanocyte inhibition and stellate scarring-like adverse effects.<sup>5</sup> To prevent this type of adverse effects we have to use the nanotechnology offering advantages like higher therapeutic efficiency, the property of drug targeting, lower toxicity, and bitter taste masking.<sup>6</sup>

### 2. Main text

### 2.1. Nanoparticles in the Treatment of skin diseases

NPs are the smallest particles, usually between 10 to 1000 nm in size.<sup>7</sup> Large number of drug molecules can entrap in NPs because of their small size and high surface area.<sup>8</sup> Different types of NPs such as metal and metal oxides, liposomes, fullerenes, nanoemulsion, polymeric NPs, solid–lipid NPs, polylactide-co-glycoside NPs, and so on, with varying physical and chemical properties, are shown in figure 1<sup>9</sup>. After appropriate surface modifications, NPs can extend circulation half-life and used to target specific diseases.<sup>10</sup> NPs can also include imaging and theragnostic agents. <sup>11,12</sup>



Figure 1: Different types of Nanoparticles

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#### 2.2. Drug delivery in skin

Barriers of skin penetration involve:

1) Stratum Corneum: The high concentration of lipids in SC's intracellular matrix, results in a hydrophobic lamellar structure that considerably enhances the barrier capabilities of SC.<sup>13,14</sup> 2) A broad dermal layer is divided into two layers of connective tissue. The superior and inferior papillary and reticular dermis. Numerous supporting elements, including hair follicles, nerves, blood arteries, perspiration and oil glands, are found in the dermis. The third one is hypodermis, that involve Blood vessels, nerves, and hair follicles. Its high fatty layer serves as an energy reservoir, it isolates the body and tolerate pressure by acting as cushions. <sup>15-18</sup> Beyond this barrier, efficient drug delivery systems are required. Nanotechnology used to modify drug penetration by controlling their release and lengthening the period of permanence by ensuring direct contact with the SC. When there is a trauma or swelling the barrier is partially disrupted. This may facilitate the penetration of NPs. NPs have the potential to improve drug particularity, bioavailability, and efficacy while also improve patient acceptance <sup>19</sup> **2.3. Introduction to Nanosponges** 

NSs are networks of sphere-shaped porous particles less than 1  $\mu$ m in diameter that are arranged in three dimensions <sup>20</sup> that serve as frames for drug molecules. Different NSs along with drugs are tabulated in **Table 1**.

Properties of NSs <sup>20, 42-45</sup> includes site-specific drug delivery with targeting and high carrier capacity. Along with swelling properties they are effectively used for low soluble drugs. NSs can trap both hydrophilic and hydrophobic moieties. They are nonlethal, porous, and stable up to temperatures of around 300°C. The nature of NSs can be either crystalline or para-crystalline and they start the controlled release of the medicine at a specified target spot <sup>41</sup>.

#### 2.3.2. Advantages and disadvantages of NSs

2.3.2.1. Advantages 46, 45, 47, 48

Due to the use of biocompatible ingredients, NSs are non-toxic, non-allergenic, or nonmutagenic. Because NSs have a diameter smaller than a bacterium (the average size of the pore is 0.25nm), the formulation is self-sterile. NSs remain stable between pH values of 1 and 11 and also aid in masking the unpleasant taste of the drug. It shows extended release for up to 12-24 hours. It is possible to convert liquids to powders that reduce drug irritability while maintaining efficacy. It is proposed to help minimize adverse effects, better stability, more elegance, and more formulation flexibility. <sup>49</sup>

# 2.3.2.2. Disadvantages 50,51

NSs include only small drug molecules. The number of crosslinking influences loading capacity of drug. There is a chance of dose dumping because the crosslinker dissolves so quickly. The particulate nature of NSs designed for topical use and the inconvenience associated with their direct application over the skin, are significant limitations.

Type of	Material	Active	Synthesis Method	Route of	Therapeutic	Purpose	Year of	References
nanosponge		Ingredient		delivery	Use		Study	
Polymer	β- cyclodextrin	Piroxicam	Melt Method	Oral	Analgesic activity	For improved internal solubility and analgesic activity	2023	21
Polymer	β- cyclodextrin	Vitamin D <sub>3</sub>	Mechanochemical method	Oral	Supplement material	To increase the chemical and biological properties of Vitamin D <sub>3</sub> .	2023	22
Polymer	β- cyclodextrin	Domperidone	Microwaveassisted approach	Oral	Antiemetic activity	For enhancing the target drug Domperidone solubility	2023	23
Biomaterial/ Metal	RNA/ ascorbic acid	Platinum, Paladinum NP	Green Synthesis	-	Hydrogen evolution reactions	Improved catalytic performance	2023	24
Biomaterial	Platelet membrane	Plateletmimicking perfluorocarbon	Ultrasonic emulsification and coextrusion	Oral	Antiplatelet activity	To solve issues of Rapid functional reversal of Antiplatelet Agents	2023	25
Polymer	β- cyclodextrin	Dithranol	Solvent evaporation and Melt method	Topical	Antipsoriatic activity	To improve poor stability and solubility	2023	26
Polymer	Ethyl cellulose	Ribociclib	Ultrasonication	Oral	Anticancer activity	Enhance Ribociclib effectiveness in the treatment of breast cancer.	2022	27

Polymer	Ethyl cellulose	Abemaciclib	Solvent evaporation	Oral	Anticancer activity	For sustained cytotoxicity against human breast cancer cells lines	2022	28
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Polymer	β- cyclodextrin	Quercetin	Ultrasound Synthesis method	Oral/ Parenteral	Anticancer activity	Increase its activity against SARS- CoV-2 virus and lung cancer	2022	29
Polymer	Ethyl Cellulose	Hespertin	Quasi-emulsion solvent diffusion	Topical	Antiinflammatory activity	Sustained anti-inflammatory effect	2022	30
Polymer	β- cyclodextrin	Limonene	Melting Method	Oral	Antibacterial activity	Solubility enhancement and stability improvement	2021	31
Polymer	β- cyclodextrin	Curcumin	Melting Method	Oral/ Topical/ parenteral	Pharmaceutical application	Examined curcumin's complexation stability	2021	32
Polymer	Ethyl cellulose	Butenafine	Solvent emulsification technology	Topical	Antifungal activity	To treat the skin infections of fungi.	2021	33
Polymer	β- cyclodextrin	Clobetasol Propionate	Melting Method	Topical	Anti-psoriatic activity	Avoid skin related side effects	2021	28
Polymer	β- cyclodextrin	Griseofulvin	Ultrasonication method	Oral	Antifungal activity	For enhancing bioavailability and masking bitter taste	2020	20

Polymer	β- cyclodextrin	Curcumin and caffiene	Melt Method	Topical	Anti-psoriatic activity	Gel is acts as a useful system for the treatment of psoriasis.	2020	34
Polymer	β- cyclodextrin	5 Fluorouracil	Ultrasound Assisted Method.	Oral	Anticancer activity	To enhance its retention in gastric tumors and to reduce its systemic side effects	2020	35
Polymer	Ethyl cellulose	Brigatinib	emulsion solvent evaporation technique	Oral	Anticancer activity	Sustained delivery for non-small lung cancer	2020	36
Polymer	Ethyl cellulose	Fluconazole	Emulsion-solvent diffusion method	Topical	Antifungal activity	To increase drug permeation through skin	2019	37
Polymer	β- cyclodextrin	Curcumin	Melting method	Oral	Anticancer activity	Study the effect on drug stability, solubility, and cytotoxicity.	2018	38
Polymer	β- cyclodextrin	Rilpivirine Hydrochloride	Microwave assisted synthesis	Oral	Anti-HIV activity	For enhancing oral bioavailability of Rilpivirine hydrochloride.	2017	39
Polymer	β- cyclodextrin	5 Flourouracil	Melting method	Oral	Anticancer activity	Enhanced anticancer activity makes medicines more soluble and limits their release.	2021	40
Polymer	Ethyl cellulose	Lornoxicam	Emulsion solvent diffusion method	Topical	NSAID activity	To reduce side effects orally and develop topical delivery	2022	41

Table 1: Drugs encapsulated in NSs

### 2.3.3. Types of NSs

There are different types (figure 2) of NS that can be designed and formulated depending on the polymer used.



**Figure 2: Types of NSs** 

### 2.3.3.1. Metal and metal oxide NSs

Metallic NSs formed by combining two or more metals and are preferable to those made of a single metal because of their porous nature. <sup>46</sup> It has a small particle size, a high surface area, and higher stability. *2.3.4.2. Silicon-based NSs* 

Silicon NS particles are made from silicon powder of metallurgical grade. Silicon NS with high porosity is used as a sensor, drug carrier material, catalyst, photosensitizer, adsorbent material, explosive material, and fuel cell electrodes. It is also used as a precursor for high surface area ceramic materials such as SiNa and SiC.<sup>10</sup>

### 2.3.4.3. Titanium-based NSs

They are prepared from a closely organized TiO<sub>2</sub> network, having higher specific area. Polystyrene microspheres coated with titanium-based NSs are constructed of polymerizable surfactants that have been copolymerized with styrene. <sup>47</sup>

### 2.3.4.4. DNAzyme NSs

DNAzymes have ease of synthesis, high selectivity, cheap cost, and high catalyst effectiveness. They can be formulated to attach and cleave a complementary target mRNA. <sup>52</sup>

### 2.3.4.5. Cyclodextrin-based NSs (CDNSs)

Cyclodextrins (CDs) are produced by enzymatic reactions on starch with Bacillus macerans amylase. The types of natural CDs ( $\alpha$ ,  $\beta$ , and  $\gamma$ ), each with glucose units of 6,7, and 8 respectively.<sup>25</sup> CDNSs are obtained by covalently bonding CDs to a crosslinker. The CD structure is a bifunctional conical cylinder with a lipophilic hole and an outer hydrophilic layer.

Sr. No.	Types of	Crosslinker used	Property	Reference
	CDNS			
1	CD-based ether NSs	Epichlorohydrin, bisphenol A diglycidyl ether, ethylene glycol diglycidyl ether, 1,4-butanediol diglycidyl ether, E-51 epoxy resin	<ul> <li>A high level of chemical resistance</li> <li>-variable swelling ability</li> </ul>	54,55
2	CD-based carbonate NSs	1,1'- Carbonyl Diimidazole, triphosgene, Dimethyl carbonate, and Diphenyl carbonate	<ul> <li>Less prone to swelling,</li> <li>Stable in mildly alkaline and acidic liquids Small surface area</li> </ul>	56,57
3	CD-based ester NSs	dianhydrides, Epiclon-B-4400; citric acid, and 2,6-naphthalene dicarboxylic acid (NDCA).	- Easily hydrolyzed in aqueous media	58-61
4	CD-based urethane/ carbamate NSs	Diisocyanates like, methylene diphenyl diisocyanate	- A rigid structure, -Resistant to chemical deterioration -Negligible swelling	62,63

<sup>53</sup> Based on the type of crosslinker used CDNS divided into various types as per Table 2. Table 2: Types of CDNSs

### 2.3.4.6. Hyper-linked polystyrene-based NSs

Large amounts of stiff intramolecular structures were added when certain polystyrene coils were suspended in diluted solvents, leading these coils to strongly constrict and form spherical NSs. In the presence of linear polystyrene non solvent, these NSs showed enhanced swelling and a rise in surface area. <sup>10</sup> 2.3.4.7. Ethyl cellulose NSs

Ethyl cellulose-based NSs involved main sustained release polymer ethyl cellulose and different crosslinkers prepared generally by an emulsion-solvent diffusion method <sup>28</sup> and an ultrasonicated emulsion solvent evaporation method. <sup>64</sup>

### **2.3.5. FACTORS AFFECTING NS FORMATION**

### 2.3.5.1. Type of polymer and crosslinker:

Whether the NS is soluble in water or any other solvent depends on the type of crosslinker that is utilised <sup>65</sup>. The cavity size of the NS must be sufficient to fit a particular drug molecule in order for complexation to occur. <sup>66</sup>

### 2.3.5.2. Degree of substitution

The capability of the parent polymer to form NS is influenced by the substituent's location, kind, and number. <sup>59</sup> If there are more substituents greater degree of crosslinking occurs thus resulting highly porous NSs. <sup>67</sup> 2.3.5.3. Drug's characteristics <sup>68</sup>

The characteristics of drugs molecules that will come into contact with NSs involve Molecular weight should be in the range of 100 to 400 Dalton. The drug structure should be less than five stacked rings. Water solubility and melting point should be less than 10mg/ml and 250°C. 50<sup>29</sup>

### 2.3.5.4. Complexation nature

It is probable that decrease in drug/NS interaction forces, such as van der Waal and hydrophobic with a rise in temperature, is the cause of the apparent stability constant of the drug NS complex decreasing with temperature.<sup>69</sup>

### 2.3.6. Loading of Drugs into NSs

Drug loading mechanism into NSs is categorized into three types. Encapsulation involves drug particles are enclosed inside the polymer. Another type involve the drug and polymers are complexed together with the aid of electrostatic charges. Conjugation includes conjugation by forming a covalent bond. <sup>70</sup> For drug loading, Plain NSs and the drug were dissolved in ultrapure water, mixed for 24 hours in the dark at a steady room temperature, and then centrifuged at 2000 rpm for a predetermined amount of time. The supernatants were then freeze-dried for 72 hours as given in Figure 3. <sup>32</sup> Then lastly collect the product and stored in well closed container.



**Figure 3: Drug Loading Method** 

### 2.3.7. Optimization of NSs

Optimization entails determining the best starting materials to create a formula that produces the required results. NSs can be easily optimized due to their simple composition, as evidenced by the examples in the table 3.<sup>46</sup>

Optimization Model	Variables	Results	Refer ence
4 <sup>2</sup> factor design	Drug: polymer ratios Volume of solvent, PVA concentrations stirring time % yield, drug content EE	The % yield increases as the drug-polymer ratio increases. Solvent volume and drug content have inverse relation.	30
Design of Expert	Polymer amount (mg) Concentration of PVA Phase evaporation rate Particle size Entrapment efficiency	Low amounts of polymer and higher levels of PVA concentration and evaporation rate could produce the lowest particle sizes. Increased polymer content provides improving drug encapsulation.	71
32 Factorial design	EC: PVA ratio (w/w) sonication time (min) Particle size Entrapment efficiency Zeta potential	Increase in the concentration of PVA, sonication time and stabilizer there is reducton in the particle size, zeta potential and increse in the stability was observed.	72
3 <sup>2</sup> factorial full design	Drug: EC ratio Stirring speed Particle size Entrapment efficiency	By increasing the stirring rate Particle size is decreased.	73
Design Expert (Robust model)	EC (mg) % PVA (w/v) Particle size % EE	Particles size increases slowly with increase in drug: polymer ratio, PVA in 0.5%. More EE results from a higher drug to polymer ratio.	74

Table 3:	Examples	of optim	ized NSs
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Plackett– Burman design	Mass of Adsorbent (mg) Sonication time (min) Volume of Eluent Ultrasonic temperature pH Vortexing time (min) Efficacy of extraction	Sorbent mass, sonication time, and eluent volume were the factors that had the greatest impact on the Extraction Recovery (%) of vitamin B3.	75
32 factorial full design	EC: PVA ratio Stirring speed (rpm) Particle size % Lemongrass Oil released	The size of the NS is reduced as the stirring rate is increased. decreased porosity and increased oil entrapment in the hydrophilic matrix, which could lead to slower release rates.	76

### 2.3.8. Preparation Method of NSs

NSs can be prepared using various methods and then loaded with varying amounts of the drug.

The various methods are given in figure.4 and are described in Table 4.



**Figure 4: Methods of Preparation of NSs** 

Method	Properties	Procedure
Ultrasoundassisted technique	Utilizes an ultrasonic water bath. It involves no solvents, making it a procedure that is good for the environment. <sup>77</sup>	Mix ingredients thoroughly before heating for five hours in a water bath to 100 °C. Washing the finished product helped to get rid of unreacted byproducts. <sup>20</sup>
Melt method	Simple, repeatable, and solventfree approach.	Polymer and cross-linker are melted together at 100°C. Using a magnetic stirrer for 5 hours, the product obtained was cooled. Soxhlet may be used to remove impurities. <sup>78</sup>
Synthesis using microwave radiation	Requires less reaction time. <sup>79</sup> Synthesises crystalline NSs. 4 times faster than the melting method. <sup>80</sup>	The drug, cross-linker, and solvent were subjected to microwave irradiation at specific conditions then the solvent was removed by distillation. Soxhlet extracts the Product for 4 hrs.
Solvent method	This method achieves a variable time range by using different temperatures. <sup>77</sup>	Following treatment with a polar aprotic solvent, an excessive amount of the polymer is added to the cross-linking agent solution, refluxing for up to 1 to 48 h. Then cool at RT. <sup>10</sup>
Emulsion solvent diffusion method(ESDM)	This process is based on the emulsification.	The internal phase consisting of a crosslinker and polymer solution, added dropwise into the aqueous solvent, stirred for 1000 rpm for 2 hrs, filtered, and dry in Oven. <sup>81</sup>
Emulsion Solvent evaporation methods	Involves the evaporation of the organic solvent.	Using ultrasonication organic phase (drug+ polymer +cross linker) was emulsified dropwise into the aqueous phase. Stir for 24 hrs, ultracentrifuge, and freeze dry. <sup>33</sup>
Quasi-emulsion solvent diffusion	Requires miscibility of solvents with water.	Emulsification of an organic drug solution that is miscible with water and contains stabilizers. 82 <sup>64</sup>
Bubble electrospinning	Process of spinning fibres with the help of electrostatic forces Geometrically observed by the size of the produced bubbles	By using 10% w/v PVA solution, Fibers /sponges were collected using a flat piece of aluminium foil placed 8 cm above the nozzle. Voltages applied ranging from 10 kV to 30 kV. <sup>83</sup>
Mechanochemical synthesis	Absorbing mechanical energy activates chemical bonds, and induces the polymer-crosslinker reaction. <sup>84</sup>	The process occurs in ball mills, with minimal use of solvent or without solvents Synthesis of NSs by mechanochemical approach involves the use of a twin-screw extruder which is capable of fine temperature control. <sup>85</sup>
Chain-Growth Polycondensation Method	polymer end group formed by the monomer reaction becomes greater in reactivity than the monomer itself. <sup>86</sup>	an initiator can be utilized to initiate polymer growth by forming high reactivity points at its tip. <sup>63</sup>

Table 4: Methods of NS preparation

2.3.10. NSs Characterization:

The various evaluation parameters of NSs are tabulated in Table 5 with details of the characterization.

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	Sr. No.	Parameters	Characterization Details
	1	Solubility studies	Higuchi and Connors demonstrate the use of phase solubility studies. <sup>87</sup> An Erlenmeyer flask is used for sampling and shaken with a mechanical shaker. centrifuged through a 3000 Dalton molecular filter. The solution analysis is done using HPLC to determine the drug concentration. <sup>88</sup>
	2	Morphology study	Scanning electron microscopy and transmission electron microscopy are used for morphology study. <sup>88</sup>
	3	Particle size, Polydispersity index, and Zeta Potential	Particle size and the PDI value indicates distribution of particles and ZP indicates particle charge. At 25°C diluted sample is used for evaluation <sup>40</sup> . Negative ZP indicates that NSs are more stable. <sup>89</sup>
	4	Thermoanalytical techniques	It is used to detect drug substance changes before thermal degradation. It may be due to evaporation and melting-like processes. <sup>90</sup> Analysis is performed in the range of 0 °C to 800 °C at a rate of 10 °C per minute in an air atmosphere with a flow rate of distinct ml/min. <sup>91</sup>
	5	X-ray diffractometry	It is used for finding the structure and the host-guest interaction of the resulting nano complexes. <sup>92</sup> The crystalline or amorphous nature of the sample can be determined at 5 to $60^{\circ}$ in a $2\theta$ scale. <sup>28</sup>
	6	Fourier transform infrared spectroscopy	To study the drug-excipients interaction and to check the physical stability of the drug. <sup>20</sup> The detection range for the detection of IR is 4000 to 650 cm- 1.72
	7	Nuclear magnetic resonance spectroscopy	For understanding the arrangement of polymers. The changes in chemical shift values ( $\delta$ ) implies proton transfer between species in the reaction and thus determines the NSs structure. <sup>93</sup>
	8	Raman spectroscopy	Raman spectroscopy explains the behavior of NSs as they change from dry to swollen. The intensity and frequency of Raman spectra indicate structural changes. <sup>94</sup>
	9	Entrapment	NSs powdered in mortar, then placed in solvent and set aside for 24 hours.
		efficiency	Then the solution was filtered and analyzed using a UV spectroscopic
			method. The formula is for the calculation of % entrapment efficiency. <sup>72</sup>
			Nanosponges actual drug content
			% Entrapment Efficiency = × 100 Therotical drug content
┢	10	Loading	For the determination of drugs loaded in NSs <sup>69</sup>
	10	efficiency (%)	
1			

Table 5: Cha	racterization	of NSs
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		After centrifugation, the supernatant was collected and analyzed with an
		ultraviolet spectrophotometer. <sup>27</sup>
		Total entrapped drug
		Loading efficiency = $\_$ × 100
		Weight of NSs
11	Thin layer	$R_f$ value results in identifying the formation of the complex between the drug
	Chromatography	and NSs. <sup>50</sup>
12	Swelling index	For swelling index determination samples are preheat-treated at 120 °C for 2
		hours. The dehydrated NS $(W_d)$ was added to the bath. After that, the outer
		surface was hydrated, and the weight $(W_h)$ was measured. <sup>95</sup> The formula for
		calculation:
		Wh
		Swelling index = $\underline{\qquad} w_d$
13.	In vitro drug	$\frac{Wh}{Swelling index} = \underline{w_d}$ Used to evaluate the kinetics and type of release of drugs. <sup>27</sup>
13.	In vitro drug release study	Wh Swelling index = $\wd$ Used to evaluate the kinetics and type of release of drugs. <sup>27</sup> For this Franz diffusion apparatus is used, consisting of the donor phase with a fixed amount of drug. The receiving phase was made up of phosphate buffer. The drug concentration was monitored using the HPLC method for 48 hours. <sup>96</sup>

# 2.3.11. Applications of NSs in the medical field and the atmosphere

2.3.11.1. Cancer Therapy

NS products are crucial in medication delivery, particularly in cancer therapy, as they are up to five times more effective at suppressing tumor development than direct injection of medicines.

# 2.3.11.2. Topical Agents

NS's application for drug delivery mechanism is a crucial technique for regulating and prolonging the release of skin-retaining medicines. To prevent rashes and other unwanted effects, an NS-based drug delivery system is used. <sup>97</sup> 2.3.11.3. NS as a Biocatalyst and Protein Delivery Carrier

Preservation of original protein structures and long-term storage are challenges for protein formulation. In the NSs-based technique, proteins are embedded in poly(amidoamine) NSs made of the CD to enhance protein stability.<sup>98</sup>

#### 2.3.11.4. NSs for Enhancement of Solubility

Using a carrier system like NSs, which helps to trap medicinal compounds in a specific pore by enhancing the solubility and bioavailability of therapeutic products in controlled release patterns, can solve the primary problem of solubility. <sup>10</sup> 2.3.11.5. NSs for Removal of Organic Impurities from Water

A well-known NS, nano porous CD polyurethanes, was employed for removing pollutants from water. The CD polymer can attract a variety of chemical molecules by forming complexes of the guest-host type.<sup>99</sup> The hydrophobic environment provided by the cavity in CD allowed it to enhance its strong attraction for organic molecules at water-solid interfaces. <sup>100</sup> The waterinsoluble CD polymers are effective in removing waste from water gathered at a particular power plant and serve an important function. <sup>101</sup>

#### 2.3.11.6 NS in SARS-CoV-2 Management

Lung type II epithelial cells or macrophage-derived NSs are capable of attracting and then removing the SARS-Cov-2 virus. The same receptors that viruses utilize to enter cells were found in the NSs, and it was believed that coronavirus would be prevented from infecting the cell by connecting with NSs. SARS-CoV-2 is neutralized and loses its capacity to infect cells when it is treated with NSs. NSs make viral mutational changes and obscure viral species. Also, it is able to identify the host cell that still serves as the virus's target and take action to neutralize it (although the quick pace of mutation will provide various difficulties for the establishment of treatments and preventative measures. In a concentration-dependent way, epithelial-NS and Macrophage NS both revealed their ability to neutralize SARS-CoV-2.<sup>102</sup> Cell membranecoated NPs called as cellular NSs. A simple way to improve SARS-CoV-2 inhibition is through surface glycan engineering of host-imitating cellular NSs.<sup>103</sup>

#### 2.3.11.7 Protective Agent against the Photo Degradation

Since gamma-oryzanol has extreme instability and photodegradation it have limited applications. Gamma-oryzanol-based nanosponges provide effective protection against the harmful effects of UV light. Using gamma oryzanol-loaded nanosponges, a gel and an O/W emulsion can be produced. <sup>104</sup>

#### 3. Conclusion

The application of NS for several purposes has created new opportunities for environmental benefits, assessment, and management. For the delivery of medications to the chosen area, the NSs act as both a variable drug transportation medium and a monitored intermediary. NSs also play a role in biosensing to detect a broad range of disease biomarkers and in the improvement of both health and the environment in many ways. Future concepts include the concurrent distribution of multiple molecules and broadening the range of applications from medicinal delivery to drug encapsulation for a variety of applications. Future trends include NSs designed for phase change material storage. Future research should be also focused on the functionalization of NSs to reduce potential toxicity, improve biosafety, and increase specificity/selectivity in biocompatibility and biodistribution. Future research should formation performances, structural concentrate on their complex

variability, commercialization, long biosafety evaluations, inexpensive/mass manufacturing, and specific nano-toxicological evaluations.

### Acknowledgements:

The management of SRES, Sanjivani College of Pharmaceutical Education and Research, Kopargaon, India, is gratefully acknowledged by the writers for providing all the necessities for the completion of this work.

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