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Artemisia: A short review on Morphology, Composition and extraction.

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# Abstract:

Artemisia species are wildly found plant across the globe. These various species grown at different climatic and demographic conditions. This review article mainly focus on morphology, chemical composition and extraction process of Artemisia species. As this plant having medicinal value, use to treat various diseases and disorders further novel research on this plant is continuously carried out.

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Article History

# Introduction:

Artemisia L. ("sweet wormwood," "annual wormwood,"), belonging to Asteraceae family of flowering plants, is annual, aromatic, herbaceous, and glabrous or sparselyhairy, with an upright brownish colored stem. It naturally grows up to a height of 1 meter, but under cultivated conditions, it may reach up to a height of 2 meters. It usually consists of a single stem with alternate branches and deeply dissected leaves. The inflorescence consists of small capitula arranged in loose panicles with bisexual disc florets at the center and pistillate ray florets at the margins.[1] Chinese phytochemist You-You Tu working in the Institute of Chinese Materia Medica, China Academy of Traditional Chinese Medicine discovered Artemisinin (Qinghaosu) in the early 1970s. [2]



## Fig. 1 various species of Artemisia [3]

The source of this medicine was the *Artemisia annua* L., which has long been used in Chinese folk medicine. This plant well-known representatives of the *Artemisia* L.genus. *Artemisia* genus is named after the Greek goddess Artemis, of the hunt and fertility. It is one of the most widely distributed genera all over the globe. More than400 species of plants distributed in various forms like grasses, shrubs, and trees. [4] As per current and future climate conditions, in mid-latitudes in western and central Europe, southeastern Asia, southeastern North, and southeastern South America are suitable areas for growth of *Artemisia*. [5] For long past years, *Artemisia* plants have been using by local peoples as remedies mainly in areas where plant occur naturally. Currently, ethnobotanical and ethno pharmacological indications of *Artemisia* genus have been proved by scientific studies. There are many species of the plant that have proven therapeutic value. [5,6]. The major antioxidant phenolic compounds foundin *Artemisia* species are Gallic acid, catechin, vanillic acid, caffeic acid, epicatechin, ferulic acid, sinapic acid, rutin, quercetin, luteolin, gentisic acid, chlorogenic acid, isoquercitrin, quercetol, kaempferol, and apigenin. [7]

#### Scientific classification and description of plant

Family: Asteraceae Subfamily: Asteroideae

Tribe: Anthemideae

Genus: Artemisia L.

The plants of this genus are annual, biennial, and perennial herbs or small shrubs and half shrubs.[8] Morphological features of the genus *Artemisia* is described as alternate leaves, small racemouse capitula, paniculate or capitate, inflorescence,

Rarely solitary; involucral bracts in few rows, receptacle flat to hemispherical, without scales and sometimes hirsute; florets all tubular, achenes obovoid, pappus absent or sometimes a small scarious ring. [9]

## **Phytochemistry:**

The phytochemical assessment of *Artemisia species* showed the presence of different phytoconstituents namely tannins, alkaloids, flavonoids, terpenoids, aminoacids, glycosides, and quinines, etc. Various secondary metabolites such as terpenoids, flavonoids, polysaccharides, and saponins have been characterized and authenticated by using High-performance liquid chromatography (HPLC), Gas Chromatography–Mass Spectroscopy (GC–MS), and Nuclear Magnetic Resonance(NMR). [10]

## Sesquiterpene lactones found in Artemisa L

Sr.	Name of constituent	Structure
No		
1.	Arablin	



Sesquiterpene lactones are formed in the plant by oxidation of the methyl part of the isopropyl group attached to the main carbon skeleton. Cyclization of farnesyldiphosphate gives (+)-germacrene A. Oxidation of the isopropenyl side chain by (+)- germacrene A-hydroxylase enzyme to primary alcohol and further oxidation by NAD (P)+- dependent dehydrogenases enzyme gives germacrene acid. This is followed by hydroxylation at the C-6 position and subsequent lactonization leads to (+)- costunolide. It is assumed that the second stage of the cyclization of

germacranolides to the guaianolide skeleton proceeds through epoxidation or hydroxylation of the costunolide. As a result of research on the homology of nucleotide sequences with known sesquiterpene monooxygenases, six promising cytochrome P450 contigs were identified and selected for functional characterization. A new cytochrome P450, cauniolide synthase, which catalyzes theformation of guaianolide cauniolide from the germacranolide substrate costunolide, has been characterized. Unlike most cytochromes P450s, cauniolide synthase has a unique mechanism of action, combining stereoselective hydroxylation of costunolide at the C-3 position with elimination, cyclization, and regioselective deprotonation. One of the promising guaianolides is arglabin, isolated for the first time from Artemisia glabella, an endemic species of Artemisia. A. glabella is a source of a number of biologically active compounds such as sesquiterpene lactones and essential oils. Arglabin is the active substance of the anticancer drug "Arglabin", which was developed at the International Research and Production Holding "Phytochemistry" and is produced on an industrial scale by the Karaganda Pharmaceutical Plant. Arglabin is found in all organs of A. glabella and throughoutthe growing season.[11]

Uther chemical constituents in various species of artemisia [12]	Other	chemical	constituents	in	various s	species	of	artemisia	[12	2]:
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Sr. No	Name	of	chemical	Structure
	constituen	ıt		
1.	Camphor			



# **Extraction technique:**

Most widely used technique to extract Artemisia species is High-pressure green technique, also called as pressurized liquid extraction (PLE) or accelerated solvent extraction (ASE) or pressurized fluid extraction (PFE). PLE is a technique characterized by the employment of suitable pressures to sustain the solvent in its

Liquid phase and high temperatures that lay below the critical point. Major advantage of this technique is, that a smaller amount of solvent is needed to carry out the process. As a result of this, the efficiency of extraction is enhanced due to a better mass transfer rate. At elevated temperature conditions, solvent solubility is improved while viscosity decreases; this helps in matrix penetration and mass transfer. So, pressure and temperature are consequential parameters to be considered when performing PLE. However, while pressure is a significant parameter, increasing pressure beyond the point necessary to maintain the solventin its liquid state has a negligible impact on the extraction process. Instead, time would be better spent on selecting a solvent that is consistent with the nature of the desired compounds. [13]

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