

<https://doi.org/10.48047/AFJBS.6.9.2024.5524-5541>



African Journal of Biological Sciences

Journal homepage: <http://www.afjbs.com>



Research Paper

Open Access

Green Chemistry Innovations for Sustainable Development

Dr. M. Sulochana

Lecturer in Chemistry SYTR Government Degree College, Madakasira. Sri Sathya Sai District, Andhra Pradesh -515301

Dr. Ram Babu

Associate Professor Department of Botany Kirori Mal College University of Delhi North Campus Delhi 110007

Dr. Shobha Thakur

Assistant professor Department of Chemistry Shuats University Prayagraj 211002

Dr. Prashant R. Mahalle,

Assistant Professor and Head, Department of Chemistry, Late B. S. Arts, Prof. N. G. Science and A. G. Commerce College, Sakharkherda

Dr. Yogesh Jorapur,

Assistant Professor, Department of Chemistry, Pratibha College of Commerce and Computer Studies, Savitribai Phule Pune University, Chinchwad, Pune 411019

Dr. Aruna Kumari Nakkella

Assistant Professor, Department of Engineering Chemistry, College of Engineering, Dr. B. R. Ambedkar University, Srikakulam

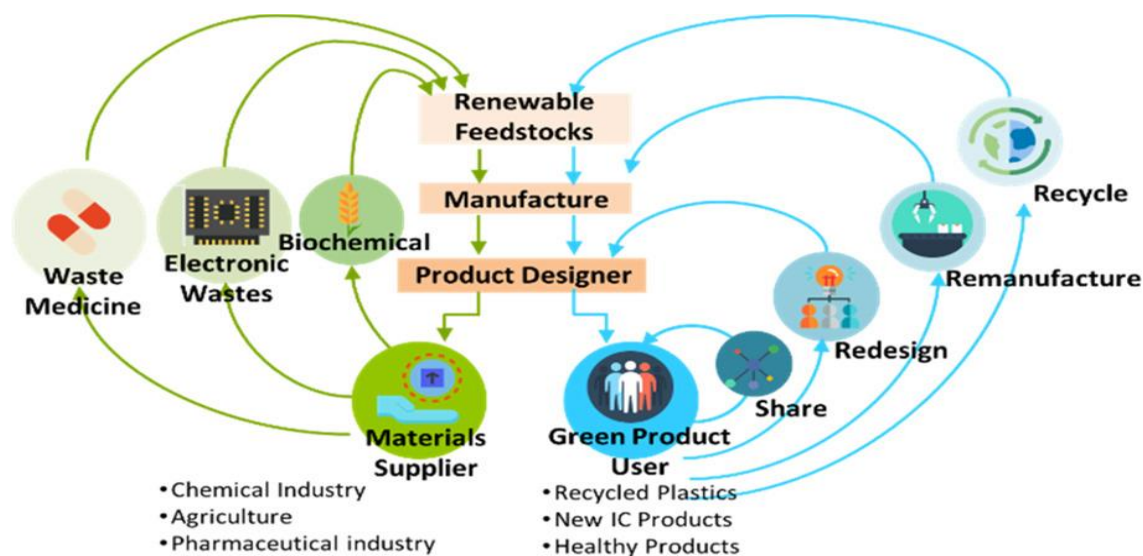


Figure 1. Graphical Abstract

Article History

Volume 6, Issue 9, 2024

Received: 10 Apr 2024

Accepted : 03 May 2024

doi: 10.48047/AFJBS.6.9.2024.5524-5541

Abstract: Green chemistry or sustainable chemistry, is a transformative approach to chemical research and industrial practices. It emphasizes the creation of products and processes that reduce the usage and production of harmful compounds. This research paper offers a comprehensive examination of recent advancements in green chemistry and their crucial contribution to the advancement of sustainable development. Given the escalating worldwide environmental concerns, it is imperative to shift from conventional chemical processes to more sustainable alternatives. This study examines different green chemistry principles, their integration into industrial operations, and their role in attaining environmental sustainability.

An important advancement in the field of green chemistry is the use of sustainable raw materials. Conventional chemical processes frequently depend on finite resources like petroleum, resulting in the exhaustion of resources and the contamination of the environment. On the other hand, renewable feedstocks obtained from biomass, such as plant materials and agricultural waste, provide a viable and environmentally-friendly option. This study explores recent progress in the research and application of renewable feedstocks, emphasizing successful examples in the manufacture of bio-based chemicals, fuels, and materials. These improvements not only decrease reliance on fossil fuels but also aid in reducing carbon footprint.

Green chemistry has made tremendous progress in the field of catalysis. Enhancing the sustainability of chemical processes can be significantly enhanced by the advancement of more effective and discerning catalysts. This paper examines current advancements in catalytic systems, encompassing both homogeneous and heterogeneous catalysts, biocatalysts, and photo-catalysts. The article discusses the utilization of earth-abundant metals, enzyme engineering, and light-driven catalysis as innovative approaches. It highlights their significance in improving reaction efficiency, decreasing energy consumption, and minimizing waste production.

Green chemistry places great emphasis on waste reduction. Conventional chemical processes can produce large quantities of dangerous waste, which presents considerable concerns to the environment and human health. This study examines novel approaches to reducing waste, such as process intensification, solvent-free processes, and the development of fundamentally safer compounds. The potential of solvent replacement techniques, such as the utilization of supercritical fluids, ionic liquids, and deep eutectic solvents, to enhance process sustainability and minimize waste is investigated.

Developing safer compounds is a crucial element of green chemistry. The formulation of compounds with reduced toxicity and enhanced biodegradability can greatly diminish their ecological footprint. This paper examines the latest progress in the development and creation of safer chemicals, which involves the utilization of computational methods to anticipate toxicity and the integration of environmentally friendly measures in chemical design. Case examples are provided to demonstrate the actual use of these concepts in the development of safer insecticides, medicines, and industrial chemicals.

This research paper offers a comprehensive analysis of innovative practices in green chemistry and their influence on the promotion of sustainable development. The study intends to stimulate additional research and adoption of green chemistry concepts by presenting successful case examples and addressing current breakthroughs. This will ultimately contribute to the worldwide endeavor to attain sustainability goals and reduce environmental consequences.

Keywords: Green chemistry, sustainable chemistry, renewable feedstocks, catalysis, waste reduction, safer chemicals, environmental sustainability, bio-based materials, energy efficiency, toxicity prediction, atom economy, green solvents, biodegradable materials, pollution prevention, eco-friendly synthesis.

1. Introduction:

The chemical industry has played a crucial role in the development of modern civilization, facilitating progress in various domains such as medicine, agriculture, materials science, and many others (1). Nevertheless, the conventional chemical techniques that have supported this advancement are now widely acknowledged as unsustainable. These operations frequently depend on finite resources, consume huge quantities of energy, and produce significant amounts of dangerous waste, which present serious threats to both the environment and human health. In light of the increasing environmental concerns such as climate change, resource depletion, and pollution, it is imperative to reconsider and revamp chemical procedures to adhere to the principles of sustainability (2).

Green chemistry, also referred to as sustainable chemistry, is a groundbreaking method that aims to tackle these difficulties. Green chemistry, as defined by Anastas and Warner, refers to the development of chemical products and processes that aim to minimize or eliminate the use and production of harmful compounds (3). This approach is not only a responsive tactic to reduce pollution after it has happened, but rather a proactive framework that aims to prevent environmental damage from the beginning. Green chemistry involves a wide array of approaches, such as utilizing renewable feedstocks, advancing catalytic processes for improved efficiency, minimizing waste generation, and designing compounds that are fundamentally safer.



Figure 2. Principles of green chemistry, from Evolution of green chemistry and its multidimensional impacts

The concepts of green chemistry present a revolutionary perspective on the future of the chemical industry. Green chemistry seeks to establish a more sustainable and resilient industrial ecosystem by giving priority to sustainability throughout the entire process of chemical production, including the selection of raw materials and the disposal of waste at the end of its useful life (4). This comprehensive strategy not only tackles environmental issues but also provides economic and social advantages, such as decreased expenses related to waste disposal and improved safety for workers and communities.

This research study seeks to investigate the most recent advancements in green chemistry and their impact on promoting sustainable development. The study aims to offer a complete overview of how green chemistry is transforming industrial practices by analyzing important areas of innovation, such as the utilization of renewable feedstocks, improvements in catalysis, waste reduction techniques, and the development of safer chemicals. In addition, the report emphasizes the significance of interdisciplinary collaboration, education, and policy support in promoting the wider acceptance of green chemistry principles.

The adoption of renewable feedstocks signifies a substantial transition from the conventional reliance on fossil fuels to the utilization of more sustainable sources of raw materials. Renewable feedstocks, obtained from biomass sources like agricultural leftovers, present a hopeful alternative that can diminish the carbon footprint of chemical operations (5). Recent progress in this field has

resulted in the creation of bio-based chemicals, fuels, and materials that are both commercially feasible and environmentally sustainable.

Catalysis, a basic component of chemical processes, has also witnessed tremendous advancements aimed at improving efficiency and sustainability. The advancement of innovative catalytic systems, including those derived from readily available metals, biocatalysts, and light-induced processes, has the capability to transform chemical production by decreasing energy usage and minimizing waste generation. These innovations play a crucial role in enhancing the sustainability and economic competitiveness of chemical processes.

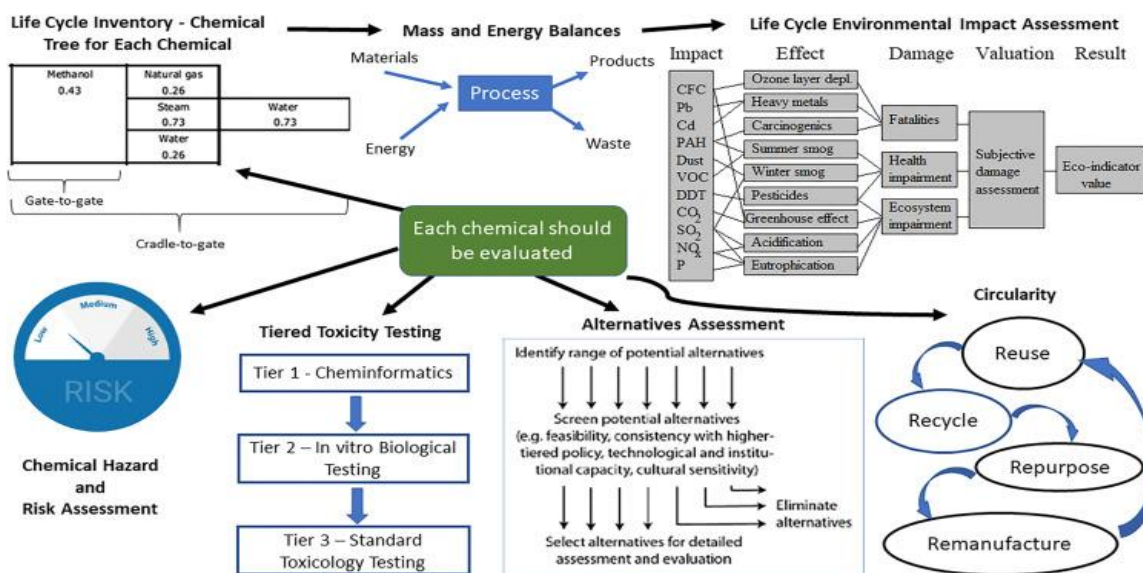


Figure 3. Systems-level view of chemical evaluation

Waste minimization is a fundamental principle of green chemistry that focuses on minimizing the environmental and health consequences of producing hazardous waste. Novel approaches such as process intensification, solvent-free reactions, and the utilization of environmentally friendly solvents such as supercritical fluids and ionic liquids are revolutionizing the management and reduction of waste in chemical processes (6). These methods not only decrease the negative effects on the environment but also enhance the effectiveness and safety of the process.

The development of safer chemicals is a fundamental aspect of green chemistry, with the goal of creating molecules that are less harmful and more capable of being broken down naturally. This facet of green chemistry guarantees that the final outcomes of chemical processes are ecologically harmless and present negligible hazards to human health. The progress in computational toxicology and green metrics is empowering chemists to create safer molecules from the beginning, hence enhancing sustainability.

2. Literature Survey

Green chemistry, or sustainable chemistry, seeks to develop chemical products and processes that minimize or eliminate the use and production of dangerous compounds. This field plays a crucial role in tackling environmental concerns and promoting sustainable development. This literature review examines recent progress in the field of green chemistry, with a specific emphasis on novel developments that promote sustainability.

2.1 Fundamentals of Green Chemistry

The formulation of twelve principles by Anastas and Warner in 1998 serves as the guiding framework for green chemistry. These principles prioritize the reduction of waste, efficient use of resources, safer chemical syntheses, energy conservation, utilization of renewable materials, and the promotion of inherently safer chemistry.

2.2 Advancements in Sustainable Chemistry

1. Catalysis

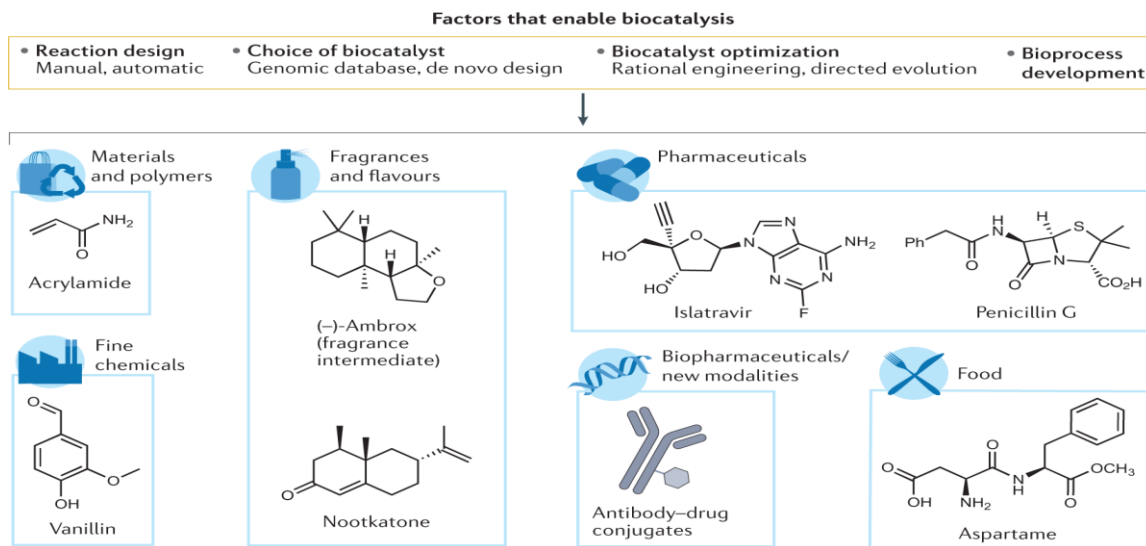


Figure 4. Bio-catalysis is enabled by various factors including reaction design, biocatalyst choice and optimization as well as bioprocess development. It has been used for the synthesis of a wide range of chemicals and pharmaceuticals, including those shown.

A. Bio-catalysis: Involves the use of natural catalysts, such as enzymes, to facilitate chemical reactions. Enzyme-catalyzed reactions generally take place under gentle conditions, resulting in reduced energy consumption and the generation of fewer byproducts (7). Recent progress has been made in developing enzymes that have been modified to have improved stability and selectivity.

B. Heterogeneous Catalysis: Solid catalysts enable reactions between substances in distinct phases. Advancements in nanocatalysts and supported catalysts have enhanced the efficiency and selectivity of reactions, resulting in less waste and energy usage.

2. Renewable Feedstocks

A. Biomass Conversion: The process of converting biomass into biofuels, biochemicals, and bioplastics has been gaining traction. Methods such as pyrolysis, hydrothermal liquefaction, and fermentation are being refined to increase productivity and improve the efficiency of the process.

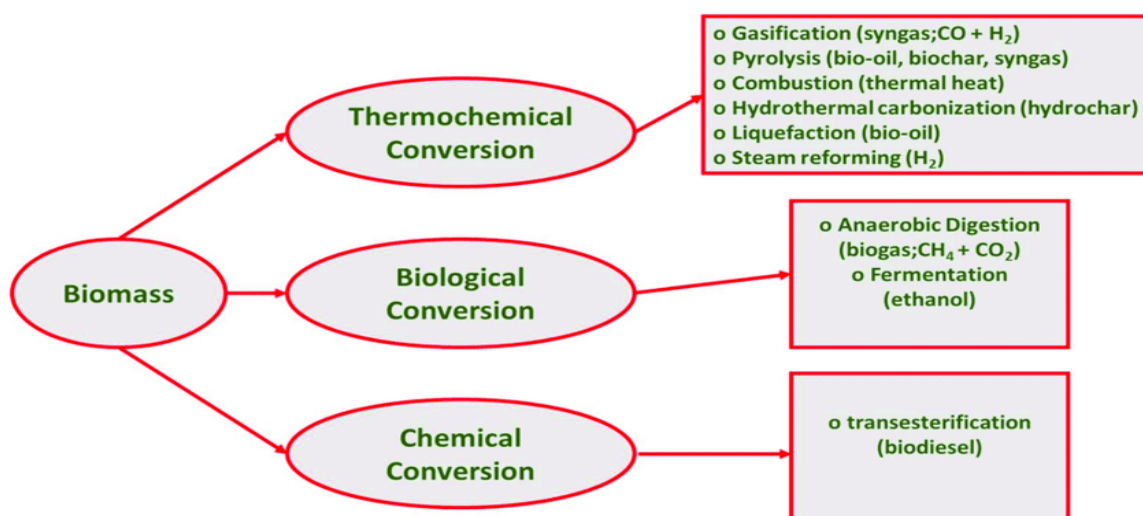


Figure 5. Biomass conversion technologies.

B. CO₂ Utilization: The process of capturing and transforming CO₂ into valuable items like fuels, chemicals, and materials shows great potential in reducing greenhouse gas emissions. The study is mostly focused on electrochemical reduction and catalytic conversion technologies.

3. Green Solvents

A. Supercritical Fluids: Supercritical carbon dioxide (CO₂) is employed as an environmentally friendly solvent for extraction, chemical synthesis, and material processing. The benefits of this substance are its lack of toxicity, non-flammability, and ease of separation from other products.

B. Ionic Liquids: These liquid substances with extremely low vapour pressure can be customized to suit specific chemical reactions. They are utilized in several applications, such as catalysis, separation processes, and electrochemistry.

3. Principles of Green Chemistry

Green chemistry, also known as sustainable chemistry, is a cutting-edge discipline in chemistry and chemical engineering that focuses on creating, advancing, and using products and procedures

Dr. M. Sulochana/ Afr.J.Bio.Sc. 6(9) (2024)

that greatly minimize or eliminate the utilization and production of harmful compounds. This interdisciplinary approach covers the complete life cycle of chemical products, starting from the extraction of raw materials and ending with the disposal at the end of their useful life (8). The goal is to reduce the environmental and health effects that are typically associated with conventional chemical activities.

The main objective of green chemistry is to develop chemical products and processes that include intrinsic safety for both human health and the environment. This entails using renewable raw materials instead of depleting limited resources, such as petroleum. Green chemists may produce more sustainable goods with a lower carbon footprint by using bio-based feedstocks. In addition, the principles of green chemistry prioritize the reduction of energy use during the chemical process. This can be accomplished by improving reaction conditions, such as using catalysts that reduce energy needs or carrying out reactions at standard temperature and pressure.

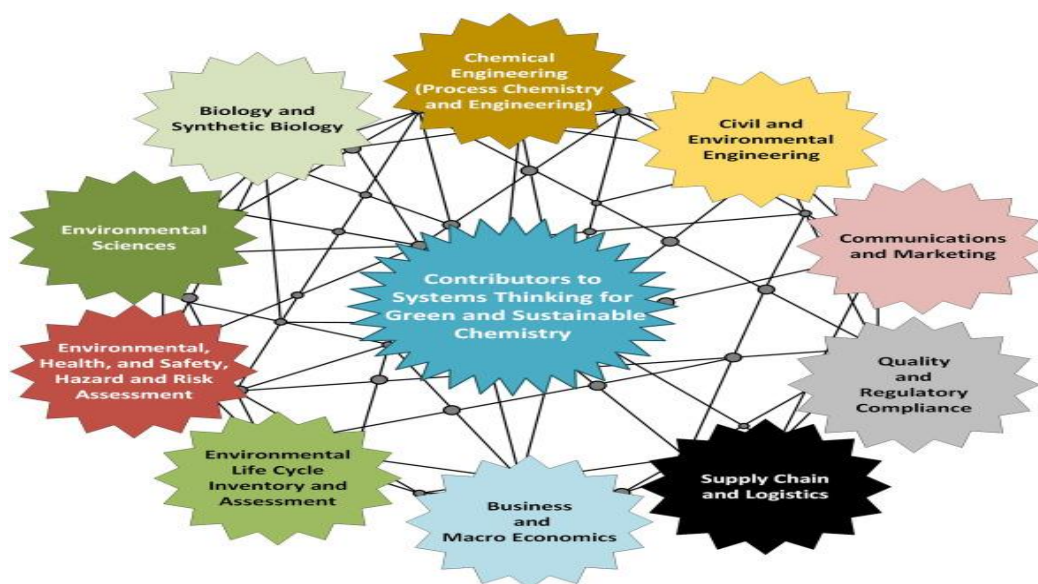


Figure 6. Contributors to systems thinking for green and sustainable chemistry

Waste reduction is a vital element of green chemistry. Conventional chemical processes frequently generate substantial quantities of waste, a considerable portion of which may pose a threat. Green chemistry aims to create processes that are highly atom-efficient, ensuring that the maximum quantity of initial ingredients is utilized in the end product, hence minimizing waste production. When trash cannot be avoided, green chemists make efforts to guarantee that it is either harmless or can be readily handled and reused.

Furthermore, green chemistry encompasses the advancement of substitute solvents and reagents that have less environmental impact. Conventional solvents, such as chlorinated hydrocarbons, possess significant toxicity and provide challenges in terms of safe disposal. Green chemistry advocates for the use of less harmful solvents, such as water, supercritical carbon dioxide, or ionic

liquids that have a less negative effect on the environment (9). These options not only mitigate the potential for exposure to harmful substances but also frequently result in enhanced efficiency and cost-effectiveness.

The concept of green chemistry also emphasizes the need of economic feasibility and social accountability. The solutions created under this framework must not only prioritize environmental sustainability but also ensure economic feasibility. This guarantees the widespread use of green chemistry in many industries, resulting in more extensive environmental advantages. Social responsibility is the evaluation of the wider consequences of chemical production and utilization, encompassing their effects on people, ecosystems, and future generations.

4. Progress in Catalysis

Catalysis is a fundamental aspect of green chemistry, acting as a catalyst to enhance the efficiency and sustainability of chemical reactions. Catalysts are chemicals that expedite chemical reactions by reducing the activation energy needed for the reaction to take place, hence increasing reaction rates without being depleted in the process (10). The capacity to selectively encourage reactions under moderate conditions is in line with the concepts of green chemistry, which prioritize atom economy, energy efficiency, and waste minimization.

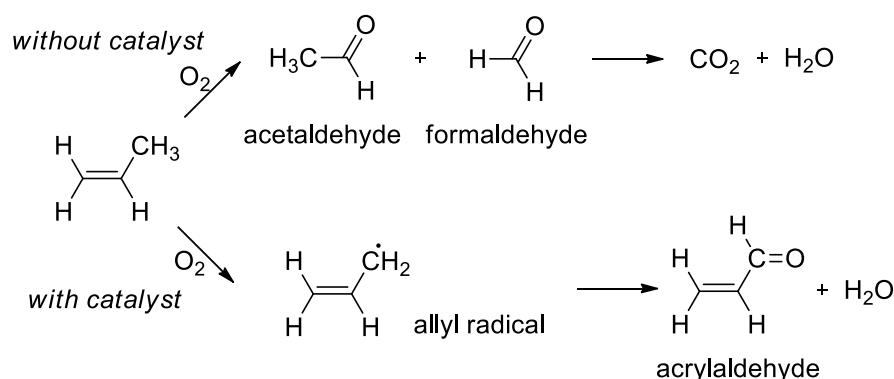


Figure 7. Reaction with and without catalysis

4.1 Differentiating Homogeneous and Heterogeneous Catalysts

Homogeneous catalysts are catalysts that exist in the same phase as the reactants, usually dissolved in the same solvent. These catalysts have a high degree of selectivity because they come into close contact with the reactants. However, their separation from the reaction mixture might be difficult, which restricts their use in industrial applications. On the other hand, heterogeneous catalysts exist in a distinct phase compared to the reactants, typically as a solid catalyst in a mixture of liquid or

gas throughout a reaction. Heterogeneous catalysts possess the advantage of being more readily separable and reusable, rendering them better suited for extensive industrial utilization.

Homogeneous catalysts	Heterogeneous catalysts
high activity	high activity
high selectivity	low selectivity
difficult separation	simple separation
low reaction temperature	high reaction temperature
easy control of mixing and concentration	difficult control of mixing and concentration
high adaptability	lower adaptability
high reproducibility	lower reproducibility
lower thermic and pressure stability	robust at high P, T

Figure 8. Difference between Homogeneous and Heterogeneous Catalysts

4.2 Biocatalysts and Enzyme Engineering

Enzymes, which are a type of biocatalysts, are becoming increasingly important in the field of green chemistry because of their exceptional selectivity, gentle reaction conditions, and capacity to be broken down naturally. Enzyme engineering approaches have been devised to customize enzymes for particular uses, improving their durability, efficiency, and specificity. These technological breakthroughs have resulted in the creation of highly efficient bio-catalytic processes used for producing medicines, fine chemicals, and biofuels.

4.3 Photocatalysis and Light-Driven Processes

Photocatalysis utilizes the energy from light to facilitate chemical reactions, providing a sustainable substitute for conventional energy-demanding procedures. Photocatalysts have the ability to initiate reactions in substrates under gentle conditions, allowing for the creation of intricate molecules with great efficiency and selectivity (11). This methodology has been implemented in diverse domains, such as organic synthesis, water treatment, and environmental remediation, demonstrating its capacity to transform green chemistry.

5. Safer Chemicals Design

Dr. M. Sulochana/ Afr.J.Bio.Sc. 6(9) (2024)

Safer chemicals design is a methodology in the realm of green chemistry that emphasizes the creation of chemical products and processes that include intrinsic safety for both human health and the environment. It entails deliberately choosing chemical components and altering chemical arrangements to decrease toxicity, flammability, and other dangerous characteristics. The objective of safer chemicals design is to minimize the hazards linked to chemical goods and processes, while preserving or enhancing their performance and functionality. By developing chemicals with inherent safety features, it is feasible to decrease the requirement for expensive and potentially detrimental mitigation measures in subsequent stages of the product's lifespan.

Safe chemical design is a core principle of green chemistry that aims to reduce the risks associated with chemical products. It entails purposefully choosing chemical components and altering chemical arrangements to decrease toxicity, flammability, and other dangerous characteristics (12). The objective of safer chemical design is to develop products that possess inherent safety for both human health and the environment, while maintaining their performance and usefulness at the same level.

An effective strategy for enhancing chemical safety is the implementation of intrinsically safer chemistry concepts. These principles prioritize the utilization of chemicals and procedures that are inherently less dangerous, such as employing less toxic substances, minimizing the usage of hazardous solvents, and decreasing the production of secondary products. By implementing these principles, chemists have the ability to create products and processes that are both safer and more sustainable.

5.1 Computational Tools for Predicting Toxicity

Computational techniques are essential in accurately forecasting the toxicity of chemicals at an early stage of the design process. Quantitative Structure-Activity Relationship (QSAR) models employ mathematical methods to establish a correlation between chemical structures and toxicological qualities (13). This enables researchers to assess the potential risks of compounds by screening them. The OECD QSAR Toolbox and ToxCast are additional tools that utilize data from chemical databases and high-throughput screening assays to forecast the toxicity of chemicals and determine their priority for subsequent testing.

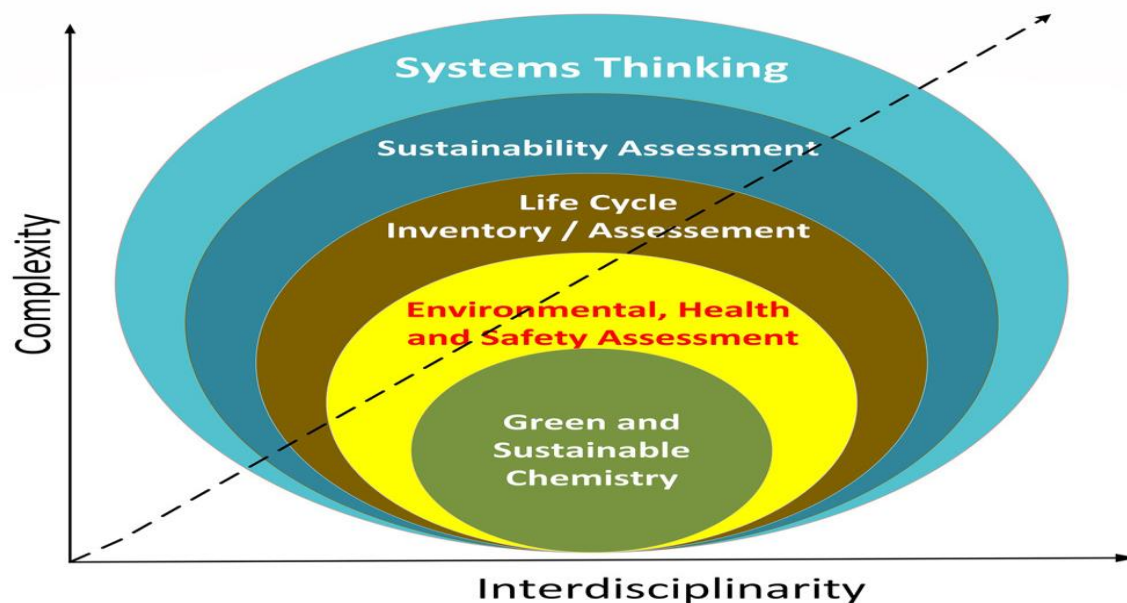


Figure 9. Environmental Impact Assessment in Chemical Design

5.2 Environmental Impact Assessment in Chemical Design

Environmental impact assessment (EIA) is a crucial element in ensuring the safer design of chemicals, as it aids in the identification and reduction of potential environmental effects that chemicals may have at every stage of their existence. Life Cycle Assessment (LCA) is a widely employed method for evaluating the ecological consequences of chemicals, taking into account variables such as resource utilization, energy expenditure, and emissions. Through the integration of EIA (Environmental Impact Assessment) into chemical design, researchers can effectively discover potential areas for reducing environmental consequences and then build products that are more sustainable.

6. The influence of Green Chemistry on different sectors

The principles of green chemistry, which prioritize sustainability, safety, and minimizing environmental damage, have significantly influenced and revolutionized multiple industries, fostering innovation and reforming conventional practices. Let's examine the impact of green chemistry on several industries:

6.1 Chemical production

Green chemistry concepts are transforming the process of chemical manufacture. Green chemistry is promoting safer and more sustainable practices by focusing on the design of chemical products and processes that minimize or eliminate the usage and production of hazardous compounds (14).

Dr. M. Sulochana/ Afr.J.Bio.Sc. 6(9) (2024)

This encompasses the advancement of more environmentally friendly methods for creating chemical compounds, the utilization of sustainable raw materials, and the minimization of both waste production and energy usage. Green chemistry involves effectively reducing the environmental impact of chemical manufacture and enhancing the safety of workers and communities.

6.2 The pharmaceutical industry

The pharmaceutical sector is progressively embracing green chemistry ideas to create safer and more sustainable ways for manufacturing drugs. Pharmaceutical businesses can mitigate their environmental footprint by employing non-harmful solvents, minimizing waste generation, and developing more efficient methods for synthesizing drugs. Green chemistry is also promoting the advancement of more environmentally friendly synthetic techniques and the utilization of renewable resources in the manufacturing of pharmaceuticals. These endeavors not only diminish the ecological impact of the pharmaceutical sector but also aid in the advancement of safer and more efficacious medications.

6.3 Agriculture and Food Production

Green chemistry is also exerting a substantial influence on agriculture and food production. Green chemistry is reducing the environmental impact of agriculture by advocating for the adoption of sustainable techniques, such as the creation of safer pesticides and fertilizers (15). Moreover, green chemistry is propelling advancements in the development of food packaging materials and processing techniques, resulting in decreased food waste and enhanced food safety. In general, green chemistry is contributing to the promotion of sustainability and environmental friendliness in agriculture and food production.

6.4 Materials Science and Engineering

Green chemistry is propelling the advancement of sustainable materials and processes in the field of materials science and engineering. Researchers are generating new possibilities for sustainable engineering solutions by developing materials that are renewable, biodegradable, and non-toxic. Green chemistry is facilitating advancements in recycling and waste management, hence diminishing the ecological consequences of material manufacturing and disposal. In general, green chemistry is contributing to the promotion of sustainability and environmental friendliness in the field of materials science and engineering.

7. Results

The study on green chemistry emphasizes its crucial role in promoting sustainability and environmental stewardship within the chemical sector. Green chemistry attempts to decrease reliance on non-renewable resources like fossil fuels and minimize the resulting environmental

effects by concentrating on the advancement and use of renewable feedstocks (16). The advancements in bio-based chemicals, fuels, and materials have exhibited encouraging outcomes, indicating that sustainable alternatives might be economically feasible and ecologically sound. These developments are made possible by the use of effective catalytic systems, which involve the use of new catalysts made from readily available metals, biocatalysts, and photocatalysts (17). These systems improve both reaction efficiency and energy savings, while also reducing waste. They adhere to the principles of green chemistry, promoting economic competitiveness and environmental protection.

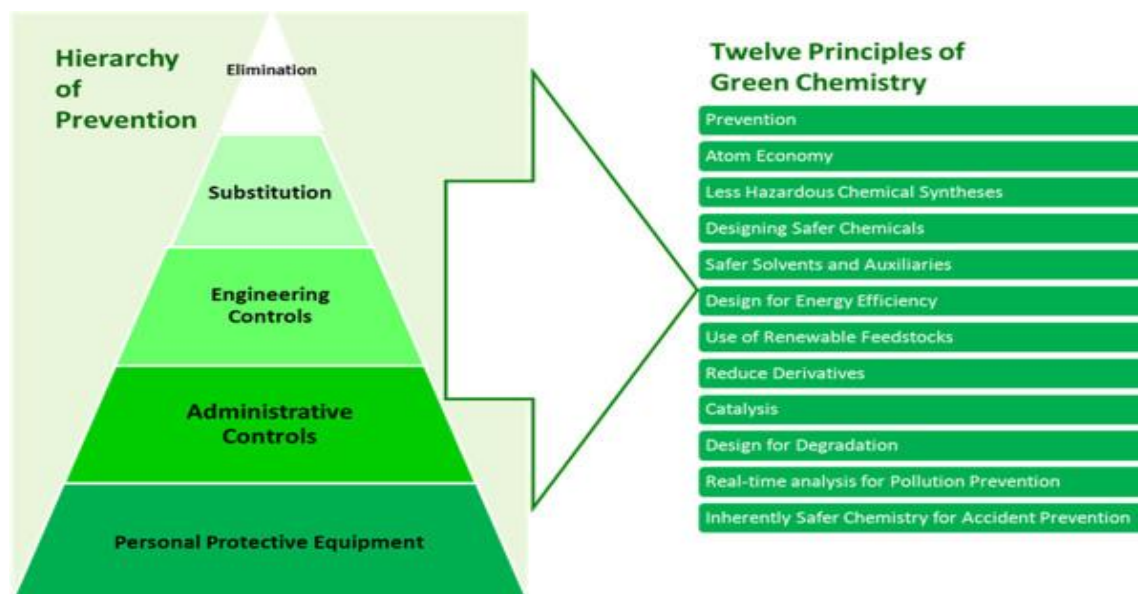


Figure 10. Green Chemistry and Application of Hierarchy of Controls.

Waste minimization is a fundamental principle of green chemistry, emphasizing the creation of processes that optimize the use of atoms and minimize the production of dangerous waste (18). Methods such as process intensification, reactions without solvents, and the use of ecologically friendly solvents such as supercritical fluids and ionic liquids have significantly transformed waste management procedures (19). These approaches not only boost the efficiency of processes but also improve safety and ensure environmental compatibility. In addition, it is crucial to prioritize the advancement of safer chemicals by focusing on the creation of compounds that possess lower toxicity, higher biodegradability, and intrinsic safety for both human health and the environment. Progress in computational toxicology and green metrics allows chemists to create safer chemicals from the beginning, therefore enhancing the overall sustainability of chemical operations.

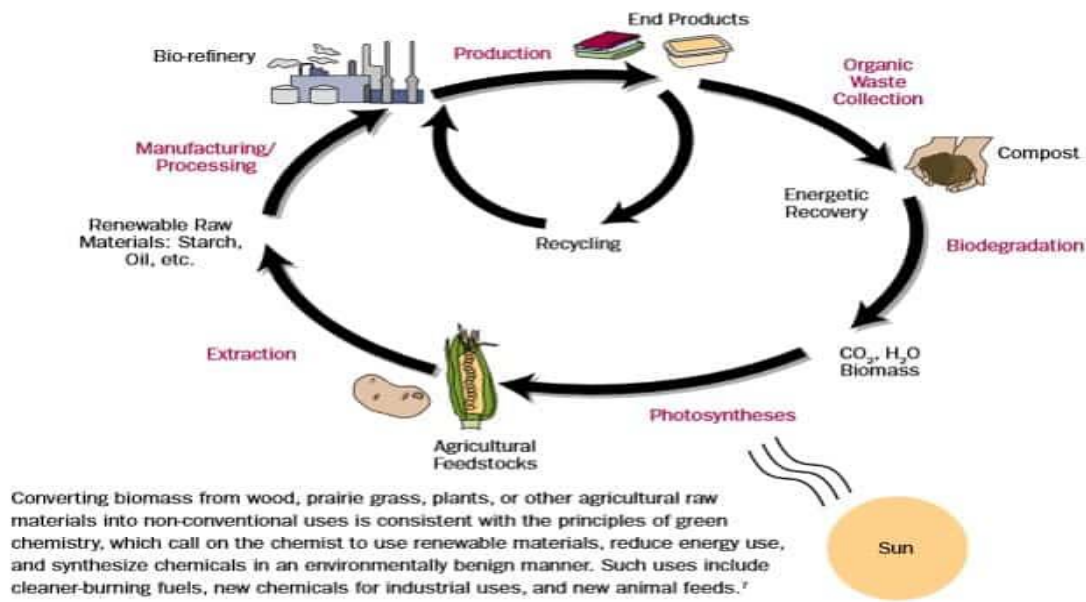


Figure 11. Green Chemistry Chart

The significance of collaboration among chemists, engineers, toxicologists, policymakers, and educators is underscored, emphasizing the multidisciplinary character of green chemistry. The collective approach is crucial for the extensive implementation of green chemistry principles and the creation of inventive resolutions to environmental obstacles. Education and policy assistance are essential for developing green chemistry and ensuring that future scientists and engineers have the required knowledge and instruments to advance sustainable practices (20). By incorporating green chemistry principles into chemical research, industrial operations, and educational programmes, we may pave the road for attaining a society that is cleaner, healthier, and more egalitarian. This encompasses the creation of novel materials that are specifically engineered to decompose naturally, techniques that minimize the release of harmful substances and the use of resources, and the fair distribution of these advancements among all segments of society. The primary objective of green chemistry is to integrate chemical manufacturing processes with natural environmental systems, promoting a sustainable future in which economic development and environmental preservation are mutually supportive.

8. Conclusion

To summarize, the discipline of green chemistry has achieved notable breakthroughs and progress that are propelling the pursuit of sustainable development. By applying green chemistry concepts, industry and researchers have successfully decreased waste, mitigated environmental harm, and created safer and more sustainable products and processes.

The importance of green chemistry in advancing sustainable development cannot be exaggerated. Industries can diminish their carbon footprint, mitigate their environmental impact, and aid in the

Dr. M. Sulochana/ Afr.J.Bio.Sc. 6(9) (2024)

conservation of natural resources for future generations by implementing green chemical practices. Green chemistry has economic advantages, including cost reductions achieved by minimizing waste and improving process efficiency.

Nevertheless, there is a significant amount of work that has to be accomplished. It is imperative for researchers, industry, and politicians to maintain their focus on green chemistry and actively work towards its wider adoption. This encompasses allocating resources towards the advancement of green technologies through research and development, fostering education and raising knowledge about the concepts of green chemistry, and enacting favorable rules and regulations.

To summarize, there is a clear and pressing call for action to be taken by researchers, industries, and politicians. It is imperative that action is taken without delay. Through the adoption of green chemistry and collaborative efforts to implement its principles, we have the potential to establish a more sustainable future for both present and future generations.

9. References

1. Dobrzański, L. A. (2006). Significance of materials science for the future development of societies. *Journal of Materials Processing Technology*, 175(1-3), 133-148. <https://doi.org/10.1016/j.jmatprotec.2005.04.003>
2. Robertson, M. (2021). *Sustainability principles and practice*. Routledge. <https://doi.org/10.4324/9780429346668>
3. Anastas, P., & Eghbali, N. (2010). Green chemistry: principles and practice. *Chemical Society Reviews*, 39(1), 301-312. <https://doi.org/10.1039/B918763B>
4. (a) Tickner, J. A., Geiser, K., & Baima, S. (2022). Transitioning the chemical industry: elements of a roadmap toward sustainable chemicals and materials. *Environment: Science and Policy for Sustainable Development*, 64(2), 22-36. <https://doi.org/10.1080/00139157.2022.2021793> (b) Jorapur, Y. R. (2022). Emerging Trends in Life Sciences. Chapter on “Ionic Liquids in Sustainable Development” (Vol. I), 211-217, Scieng Publications.
5. Maji, S., Dwivedi, D.H., Singh, N., Kishor, S., Gond, M. (2020). Agricultural Waste: Its Impact on Environment and Management Approaches. In: Bharagava, R. (eds) *Emerging Eco-friendly Green Technologies for Wastewater Treatment. Microorganisms for Sustainability*, vol 18. Springer, Singapore. https://doi.org/10.1007/978-981-15-1390-9_15
6. (a) Grillo, G., Cintas, P., Colia, M., Calcio Gaudino, E., & Cravotto, G. (2022). Process intensification in continuous flow organic synthesis with enabling and hybrid technologies. *Frontiers in Chemical Engineering*, 4, 966451.

- <https://doi.org/10.3389/fceng.2022.966451> (b) Jorapur, Y. R.; Chi, D. Y. (2011). Intramolecular Cycloalkylation of Pyrrole in Ionic Liquids and Immobilized Ionic liquids. *Bull. Korean Chem. Soc.* 32, 3130-3132. <https://doi.10.5012/bkcs.2011.32.8.3130> (c) Jorapur, Y.; Chi, D. Y. (2023). Synthesis of a Precursor 1,2-Bis(1H-pyrrol-2-yl)ethane Towards Novel Pyrrole Macrocycles in Ionic Liquids as a Green Approach. *Eur. Chem. Bull.* 12 (6), 4738–4747. <https://doi:10.31838/ecb/2023.12.si6.417>
7. Jegannathan, K. R., & Nielsen, P. H. (2013). Environmental assessment of enzyme use in industrial production—a literature review. *Journal of cleaner production*, 42, 228-240. <https://doi.org/10.1016/j.jclepro.2012.11.005>
 8. Keoleian, G. A., & Menerey, D. (1994). Sustainable development by design: review of life cycle design and related approaches. *Air & Waste*, 44(5), 645-668. <https://doi.org/10.1080/1073161X.1994.10467269>
 9. (a) Ivanković, A., Dronjić, A., Bevanda, A. M., & Talić, S. (2017). Review of 12 principles of green chemistry in practice. *International Journal of Sustainable and Green Energy*, 6(3), 39-48. doi: 10.11648/j.ijrse.20170603.12 (b) Advani, K. R.; Jorapur, Y. R.; Thomas, A. B. (2021). Fischer Indole Synthesis Catalyzed by Phosphomolybdic Acid in Poly(ethylene glycol) (400) as Recyclable Reaction Medium. *Vidyabharati Intl Interdiscip. Res. J.* 1, 54-59. (c) Jorapur, Y. R.; Chi, D. Y. (2006). Ionic Liquids: An Environmentally Friendly Media for Nucleophilic Substitution Reactions. *Bull. Korean Chem. Soc.* 27, 345-354. DOI:10.5012/bkcs.2006.27.3.345
 10. Atadashi, I. M., Aroua, M. K., Aziz, A. A., & Sulaiman, N. M. N. (2013). The effects of catalysts in biodiesel production: A review. *Journal of industrial and engineering chemistry*, 19(1), 14-26. <https://doi.org/10.1016/j.jiec.2012.07.009>
 11. Lang, X., Chen, X., & Zhao, J. (2014). Heterogeneous visible light photocatalysis for selective organic transformations. *Chemical Society Reviews*, 43(1), 473-486. <https://doi.org/10.1039/C3CS60188A>
 12. Carson, P. A. (2002). *Hazardous chemicals handbook*. Elsevier.
 13. Perkins, R., Fang, H., Tong, W., & Welsh, W. J. (2003). Quantitative structure-activity relationship methods: Perspectives on drug discovery and toxicology. *Environmental Toxicology and Chemistry: An International Journal*, 22(8), 1666-1679. <https://doi.org/10.1897/01-171>
 14. Abdussalam-Mohammed, W., Ali, A. Q., & Errayes, A. O. (2020). Green chemistry: principles, applications, and disadvantages. *Chem. Methodol*, 4(4), 408-423. <https://doi.org/10.33945/SAMI/CHEMM.2020.4.4>

15. Fenibo, E. O., Ijoma, G. N., Nurmahomed, W., & Matambo, T. (2022). The potential and green chemistry attributes of biopesticides for sustainable agriculture. *Sustainability*, 14(21), 14417. <https://doi.org/10.3390/su142114417>
16. Sheldon, R. A. (2016). Engineering a more sustainable world through catalysis and green chemistry. *Journal of the Royal Society Interface*, 13(116), 20160087. <https://doi.org/10.1098/rsif.2016.0087>
17. Rodríguez-Padrón, D., Puente-Santiago, A. R., Balu, A. M., Muñoz-Batista, M. J., & Luque, R. (2019). Environmental catalysis: present and future. *ChemCatChem*, 11(1), 18-38. <https://doi.org/10.1002/cctc.201801248>
18. Doble, M., Rollins, K., & Kumar, A. (2010). *Green chemistry and engineering*. Academic Press.
19. Hessel, V., Tran, N. N., Asrami, M. R., Tran, Q. D., Long, N. V. D., Escribà-Gelonch, M., Tejada, J.O., Linke, S. & Sundmacher, K. (2022). Sustainability of green solvents—review and perspective. *Green Chemistry*, 24(2), 410-437. <https://doi.org/10.1039/D1GC03662A>
20. García-Serna, J., Pérez-Barrigón, L., & Cocero, M. J. (2007). New trends for design towards sustainability in chemical engineering: Green engineering. *Chemical Engineering Journal*, 133(1-3), 7-30. <https://doi.org/10.1016/j.cej.2007.02.028>