https://doi.org/10.48047/AFJBS.6.13.2024.7588-7608



General Approach to Nanocomposites Preparation

Laxmi Yadav *, Swarnakshi Upadhyay,

School of Pharmaceutical Science, CSJMU, Kanpur-208024, UP, India

Volume 6, Issue 13, Aug 2024

Received: 15 June 2024

Accepted: 25 July 2024

Published: 15 Aug 2024

doi: 10.48047/AFJBS.6.13.2024.7588-7608

ABSTRACT

Nanocomposites are a form of nanomaterial consisting of one or more nanoscale phases embedded in a ceramic matrix, metal, or polymer matrix.. The structure of traditional particle-reinforced composites is drastically replaced by the nanostructure created by adding discrete nanoparticles into a continuous solid matrix.. The author of this article discusses nanocomposites, including their structure, classification, preparation process, applications, and potential future development. Over the course of the last twenty years, nanocomposites have become a novel class of materials that have drawn a lot of attention from researchers and engineers. When nanoparticles are less than 5 nm in size, catalytic activity in nanocomposite systems is triggered; when they are less than 20 nm, hard magnetic materials are softened; when they are less than 50 nm, refractive index changes are produced; when they are less than 100 nm, superparamagnetism and other electromagnetic phenomena are produced; and when they are less than 100 nm in the same range, strengthening and toughening are enhanced. Following the recent discovery of carbon nanotubes (CNTs), a novel and intriguing property has been introduced to this field, opening the door to the possibility of using CNTs to produce nanocomposites with improved thermal, electrical, and mechanical characteristics. Based on the matrix materials, nanocomposite materials can be categorized into three groups: Metal Matrix Nanocomposites (MMNC), Polymer Matrix Nanocomposites (PMNC), and Ceramic Matrix Nanocomposites (CMNC). Keywords: Nanomaterials, nanocomposites, solid matrix, and resources

Keywords: Nanomaterials, nanocomposites, solid matrix, and resources derived from fossil fuels, renewable resources, biobased materials, biomasses chemistry, and sustainability.

INTRODUCTION

In contrast to micro composites, a nano-composite is a multiphase material where the distances between the composite phases are nanoscale, or phase with a size less than 100 nm exists. Weak interactions, like vanderwaals, can be used to combine two materials to create a composite nanomaterial. Arbitrary inorganic units of nano composite materials include carbon nanotubes (CNTs), nanoparticles, nanorods, and nano fibers. Weak electrostatic interactions and hydrogen bonding can also be produced by covalent bonds. Several nanocomposites' constituent parts work

together to produce synergistic properties. The multifunctional features of the nanocomposites include strong mechanical properties, high electrical conductivity, and high surface to volume ratio, high redox reactivity, and catalytic activity for loading biomolecules such as enzymes. Nanocomposites find application as transducer materialism bio-nano electronic devices that use gas sensing, batteries, pathogen and cancer cell detection, targeted medication delivery imaging, and artificial implants An major distinction between nanocomposites and conventional composites is the increased interfacial area per volume of material between the nanoparticle and matrix that results from a higher specific number of nanoparticles. This will therefore affect the properties of nanocomposites. With a projected 25% annual growth rate, the fastest-growing applications for nanocomposites are expected to be in elastomers and engineering plastics. Their potential is so great that they are useful for a variety of applications, from packaging to biomedical ones.[1] High-performance materials with unusual property combinations and exceptional design opportunities are known as nanocomposites. Their potential is so great that they may be applied to a variety of fields, from packaging will biomedical applications, with the engineering plastics and elastomers industry expected to have the fastest annual growth rate. Applications of nanocomposites thus provide new technical and commercial options for prospective parts of the aerospace, automotive, electronics, and biotechnology industries, as they are environmentally friendly[2]

According to reports, in nanocomposite systems, catalytic activity is triggered in particular when particles are smaller than 5 nm; harder magnetic materials are softened when particles are smaller than 20 nm; refractive index changes are produced when particles are smaller than 50 nm; superparamagnetism and other electromagnetic phenomena are produced when particles are smaller than 100 nm; and strengthening and toughening are enhanced when particles are smaller than 100 nm within the same range.[3]

HISTORICAL PERSPECTIVES

In 1961, Blumstein released the first literature review on nanocompositesHis further research in 1965 additionally confirmed the increased thermal stability His further research in 1965 additionally confirmed the increased thermal stability "nanocomposites" was coined in 1970 by Theng. As early as 1976, polyamide nanocomposites were published, and nanocomposites were originally mentioned in literature in 1950. However, nanocomposites were not extensively

explored in academic and industrial facilities prior to Toyota researchers beginning a comprehensive examination of composites made of polymer and layered silicate clay materials. As stated in Assonant's 2009 explanation, Toyota "Timing belt covers for Toyota cars were made using clay/nylon-6 nanocomposites for the first time in 1990, which launched the field of polymer nanocomposites. By examining their applications, advantages, and disadvantages, raw materials and manufacturing techniques that are most suitable for producing a given nanomaterial are identified in the process of developing nanocomposites. The use of nanocomposites has a long history. Clay and straw have been combined in Egypt to create bricks. The Mongols employed composites in their battle, and even more recently, during World War II, materials based on composites were employed in military equipment. In the present period, a vast array of composites is employed in various disciplines. The advent of the scanning tunneling microscope in 1981 marked the beginning of modern nanotechnology since,. Characterization allowed scientists and engineers to see and work with individual atoms Current advancements in the synthesis, and manipulation of materials at the nanoscale have enhanced their application as fillers in novel kinds of nanocomposites.[4]

Since the 1990s, attention has been drawn to nanocomposite systems, particularly CNTreinforced nanocomposites, and it has been noted that there has been a consistent and ongoing increase in the amount of publications on this subject, including reviews.[5,6]

A small quantity of nanofillers loading significantly improved the Nylon-6 Nanocomposites mechanical and thermal properties, according to research done in the 1990s by Toyota Central Research Laboratories in Japan. According to Kanartzidis, the morphology and interfacial characteristics of nanocomposite materials, in addition to the attributes of their separate parents (in this example, nano filler and nylon), determine their properties [7].

In summary, the field of nanocomposites has attracted a lot of attention because of the special combination of nanomaterial characteristics, such as size, low concentrations, and mechanical properties, which are actually necessary to effect change in a polymer matrix. These characteristics are coupled with the advanced characterization and simulation algorithms that are now available. Furthermore, a lot of polymer nanocomposites are particularly appealing from a production perspective since they can be assembled and processed similarly to traditional polymer composites[8]

Chemistry in Britain released an article titled "Nano sandwiches" in 1998, which claimed that "Nature is an outstanding chemist with incredible talent." [9]

STRUCTURE:

Nanosized reinforcement components, such as particles, whiskers, fibers, nanotubes, etc., are typically found in the matrix material of nanocomposites. Quantitative assessment of crystalline and nanostructures in nanocomposites is accomplished through a variety of tools and methods, including differential scanning Calorimetry (DSC), Fourier transformed infrared microscopy (FTIR), scanning tunneling microscopy (STM), X-ray diffractometry (XRD), scanning and transmission electron microscopy (SEM/TEM), simultaneous small angle X-ray (SAXS), and nuclear magnetic resonance (NMR). Nanocomposites are typically made up of a polymer and a nanoparticle. Based on their geometric properties, nanomaterials such as nanoparticles, nanotubes, nanofibers, fullerenes, and nanowires are categorized into Particle, layered, and fibrous material are the three classes. While silica nanoparticles, carbon black, and Polyhedral Oligomer Silsesquioxanes (POSS) can be classified as nanoparticle reinforcing agents, carbon nanotubes and nanofibers are examples of fibrous materials. Like organ silicate, the filler is regarded as a layered nanomaterial, when it has a high aspect ratio (30-1000) plate-like structure and a thickness of nanometers This new generation of composite materials presents many opportunities for novel approaches in their manufacturing, characterization, analysis, and modeling due to the change in length scales from meters (finished woven composite parts), micrometers (fiber diameter), and submicrometers (fiber/matrix interphase) to nanometers (nanotube diameter). [10]

Features of Nanocomposite Materials: The properties of the nanofillers and polymer, the structure of the composite that is formed, and the combination of clay and polymer all influence the nanocomposites' properties. For a different physical attribute, a nanocomposites' ideal structure might not be the best.

Mechanical characteristics: Tensile strength, elongation, and modulus are just a few of the mechanical qualities of nanocomposites that are influenced by the surface morphology and the material employed in their manufacture. Together with the high stiffness and high aspect ratio of

the nanofillers, the good affinity between the polymer and the nanofillers is responsible for the improvement of the mechanical properties of polymer nanocomposites.

Thermal characteristics: DSC can be used to examine the thermal characteristics of nanocomposites. The thermal stability of the nanocomposites can be computed based on the weight loss that occurs after heating. The HDT can be used to measure the heat resistance of nanocomposites under external loading. Numerous scholars have looked into how clay content affects HDT. High-performance thermal management systems, heat sinks, connections, printed circuit boards, and thermal interface materials are just a few uses for nanocomposites with good thermal conductivity.

Electrical characteristics: A number of variables, including aspect ratio, dispersion, and alignment of the conductive nanofillers inside the structure, affect a nanocomposites' electrical properties. High energy densities and low driving voltages are two of the outstanding electrical features of the CNT-containing nanocomposites. The ether/clay nanocomposites that have undergone organic modification display ionic conductivity thais significantly more than that of similar clay. With a very little loading (0.1 wt% or less) of nanotubes in the nanocomposites, the materials' conductivity rose by several orders of magnitude without impacting other properties like optical clarity, mechanical capabilities, and low melt flow viscosities. • Electrostatic dissolving, electrostatic creating artwork, electromagnetic radiation shielding, printed circuit wiring, and transparent conductive coating are just a few of the numerous domains in which conductive nanotechnology has found use.

Titanium dioxide: metal quantum dots, and biologically modified materials have optical characteristics. Among the nanomaterials with outstanding optical qualities for real-world uses like solar cells, sensors, coatings, etc. are nano clays. Because of their optical characteristics, nanomaterials are used in sensors such as electrooptical, photo luminescent, nanostructured surface Plasmon resonance, and fiber-optic sensors.

Barrier qualities: Due to their high aspect ratio and ability to create a convoluted channel that slows the passage of gas molecules through the matrix resin, the nanocomposites have very good barrier capabilities against gases. The filler creates a convoluted pathway inside the nanocomposites structure that diffusing penetrants must follow. The longer diffusive path that the penetrants have to take when there is filler results in decreased permeability. The polyimide

nanocomposites with a tiny amount of layered silicate show barrier properties against vapors of ethyl acetate and small gases like oxygen, carbon dioxide, helium, and nitrogen.

Rheological characteristics: PCL/nylon 6 nanocomposites exhibited a substantially different flow behavior than the comparable pristine matrices. The thermo-rheological characteristics of nanocomposites are determined by the behavior of matrices. When it comes to composite processing, composite dynamics, and microstructure characterization, the viscoelastic properties of nanocomposites are crucial.

Factors impacting the properties of nanocomposites: Several elements impact how well nanocomposites perform, including. [11]

- 1. Volume fraction of nanoparticles.
- 2. The nanoparticles characteristics.
- 3. Dimensions and form of the implanted substance.
- 4. The method by which nanocomposites are made.
- 5. Level of Blending.
- 6. Aggregate formation.
- 7. Even distribution of the two stages.
- 8. The reinforced material and matrix interface is present.
- 9. Interface bonding involving the matrix and reinforced material
- 10. The nanoparticles surface area.

11. Spread.

1. CLASSIFICATION:

The discovery of carbon nanotubes, or CNTs, in 1991 [3] and their subsequent application to create composite materials with some of the special mechanical, electrical, and thermal properties associated with CNTs have brought a new and intriguing aspect to this field [4, 6]. Three categories, namely Ceramic Matrix Nanocomposites (CMNC), Metal Matrix Nanocomposites (MMNC), and Polymer Matrix Nanocomposites (PMNC), can be used to classify nanocomposite materials based on their matrix components. The two categories of

nanocomposites that exist are matrix-based and reinforcement-based. are divided, and there are further subcategories within these two groups. [12]

Metal/metal nanocomposites: Bimetallic nanoparticles with a wide range of catalytic and article properties will be employed as alloys or as shells. Chemical and biosensors employ metal oxide nanoparticles and nanowires with semiconducting characteristics as fillers and gas sensing materials. Semiconductor metal oxides are less expensive, easily distributed, and stable in air.

Nanocomposites made of metal and ceramic:

These materials have enhanced mechanical, electrical, magnetic, and chemical capabilities Metal nanoparticles can be coated on a ceramic support by evaporating metal on the selected substrate metal nanoparticles, or they can be distributed using solvent chemistry. Novel techniques for processing complex nanocomposites, such as electrospinning, electrochemical scanning probe approaches, and template synthesis.

Ceramic/ceramic nanocomposites: To solve fracture failure concerns, these nanocomposites are used in artificial joint implants. Allowing patients to move around more freely and avoiding the need for expensive surgery. For instance, alumina-toughened zirconia. Types of polymer-based nanocomposites include the following:

1. Polymer/ceramic nanocomposites: These comprise uniformly dispersed single ceramic layers, each one a nm thickness, forming a continuous matrix. Because of their dipole-dipole interaction, ceramic layers align themselves parallel to one another. Because ceramic powder has a high dielectric constant and is made of extremely flexible polymers, polymer-ceramic nanocomposites, which are made using ceramic nanopowder and polymer matrices, have advantages when it comes to embedding capacitors. The degree of cross-linking determines the mobility of the metal atoms Because of their improved dielectric constant One potential material for embedded capacitors is polymer/ceramic nanocomposites, which are polymer matrices filled with ceramic nanopowder .[13]

2. Inorganic/organic polymer nanocomposites: These nanocomposites consist of metal clusters ranging in size from 1 to 10 nm, which are incorporated into a polymer matrix. The mobility of metal atoms on the polymer surface is determined by the size and structure of the

nanocomposites. For instance, in polymethyl methacrylate (PMMA) polymers, the quantity of cross-linking determines the cluster size.

3. Inorganic/organic hybrid nanocomposite: These are heterogeneous nanocomposites, which are homogenous systems comprising monomers and miscible organic/inorganic components.

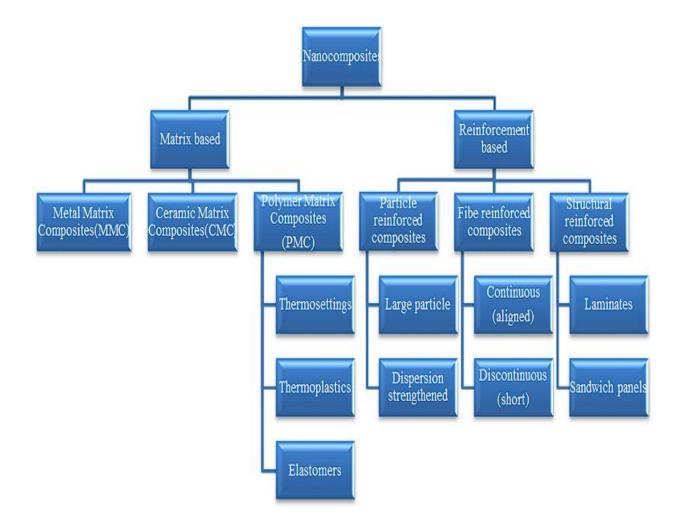
4. Polymer/layered silicate (PLS) nanocomposites: Compared to virgin polymer and traditional composites, PLS nanocomposites have exceptional characteristics.

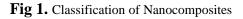
5. **Polymer/polymer nanocomposites**: Polymers are quite ever struggling to be chip and offered property profiles. Block co-polymer self-assembly and offer nanostructured plastic with as-yet-undiscovered combinations of attributes are getting closer to each other. Even when different polymer mixes are homogeneous, they frequently phase segregate.

6. **Bio-composites:** Orthopedics, dentistry, and other load-bearing applications employ metals and metal alloys. Collagen comes in a variety of forms and is rather plentiful

7. Nanocomposites based on carbon nanotubes: Carbon nanotubes are used in composite materials, chemical and biosensors, nano electronics devices, and composites because of their mechanical and electrical characteristics. They are conductive by nature, and some of the components that use them call for the electrostatic potentials to be released. These electrically conductive carbon nanotube-based nanocomposites are appropriate for uses where the capacity to release electrostatic potentials is necessary. Single-walled nanotubes (SWNTs) and multiwall nanotubes (MWNTs) are the two varieties of carbon nanotubes: With a diameter ranging from 1mm to 3nm, a cylindrical single graphite nanostructure sheet is rolled to make SWNTs, whereas MWNTs are made up of concentric single nanotubes arranged in a coaxial arrangement.

8. Nobel metal-based nanocomposites: Melted or dissolved metal nanoparticles are combined with a polymeric matrix. High sensitivity porous metal oxide nanocomposites with easily adjustable porosity, low-temperature encapsulation, great chemical stability, little mechanical and biodegradable stability, and low preparation temperature are all attributes of these nanocomposites finding of oxidizing and reducing swell mechanical and biodegradable stability, and low preparation temperature finding of oxidizing and reducing swell mechanical and biodegradable stability, and low preparation temperature are all attributes of these nanocomposites finding of oxidizing and reducing swell mechanical and biodegradable stability, and low preparation temperature are all attributes of these nanocomposites finding of oxidizing and reducing swell mechanical and biodegradable stability, and low preparation temperature are all attributes of these nanocomposites finding of oxidizing and reducing.





Manufacturing of nanocomposites:[15]

It is possible to create these nanocomposites mechanically or chemically. There are three different kinds.

- 1. Nanocomposites with a ceramic matrix (CMNC)
- 2. MNCs, or metal nanocomposites
- 3. PMNCs or polymer nanocomposites
 - 1. CMNCs, or ceramic matrix nanocomposites:

The powder process: Raw materials (mostly powders with small average sizes, homogeneity, and high purity) are combined with organic or aqueous media and dried using lamps, ovens, or freeze drying through the use of wet ball milling or attrition milling processes. Subsequently, the solid material is cemented by injection molding, pressure filtering, slip casting, hot pressing, and pressure-gas sintering.

Advantages: simple Restrictions include agglomeration, limited phase dispersion, high temperature, low formation rate, and secondary phase development in the final product.

Polymer precursor process: Using a microwave oven to pyrolyze a combination including a Si-polymeric precursor and matrix material to produce reinforcing particles.
Benefits include: the ability to prepare the final particles and improved dispersion of reinforcement.

Limitations: because of the agglomeration and dispersion of ultra-fine particles, limitations exist in homogeneous and phase-segregated materials.

Sol-Gel process: An organic molecule precursor dissolved in organic medium undergoes polycondensation and hydrolysis processes. Metal-oxygen connections in three-dimensional polymers (sol-gel) are created as a result of reactions, which are then dried to produce a solid and further solidified by heat treatment.

- **Benefits include:** possessing a high degree of chemical homogeneity, being versatile, controlled stoichiometry, yielding high-purity products, and generating metal-oxygen links to build three-dimensional polymers. A single matrix or more. specifically useful for creating composite materials using liquids or viscous fluids.
- Limitations: Compared to the mixing process, there is more shrinkage and less voids.[16]
- 1. Metal nanocomposites:

Spray paralysis: This process involves dissolving the inorganic procedures (beginning materials) in an appropriate solvent to create a liquid source, utilizing an ultrasonic atomizer to create a mist from this liquid source, transporting the mist into a heated chamber via a carrier gas, vaporizing the droplets inside the chamber, filtering them, letting the droplets break down to create the right oxide materials, and then reducing the metal oxides only to make the matching metallic materials.

Benefits include: reproducing size and quality as well as the efficient production throughout multicomponent systems of ultra-fine, spherical, and homogeneous powder.

Liquid infiltration: The matrix metal material is combined with fine reinforcing particles to encourage consolidation and eliminate internal porosity and thermally treated below the matrix melting point. This causes the matrix to melt and envelop the reinforcements.

Benefits include: quick solidification, quick contact times between the reinforcements and matrix, molding into various and nearly-net shapes with varying stiffness and improved wear resistance, and production on both an industrial and lab scale.

Limitations: high temperature application, reinforcing segregation, and the production of undesirable byproducts during processing.

The Rapid The process of solidification Process (RSP): entails melting the constituent metal parts jointly and ensuring homogeneity by maintaining the melt above the critical line of the miscibility gap between the components. The melt can be rapidly solidified by several processes, including melt spinning.

Benefits: easy to use and efficient.

Limitations: non-homogeneous fine particle distribution, induced agglomeration, and only metal-metal nanocomposites.

RSP with ultrasonic: This technique uses ultrasonic to improve wettability and mix the matrix and reinforcements.

Advantages: even with small particles, there is good dispersion without agglomeration. High energy ball milling: This method involves grinding particles together to create nanocomposites, or the necessary nanosized alloy.

Benefits: consistent dispersion and homogenous mixing.

PVD/CVD: PVD is the sputtering/evaporation of various components to create a vapor phase; supersaturation of the vapor phase in an inert atmosphere encourages the condensation of metal nanoparticles; thermal treatment in an inert atmosphere consolidates the nanocomposites. CVD involves using chemical reactions to produce material vapors, which are then condensed.

Benefits include the capacity to generate materials that are extremely pure and dense, thick films

that are homogeneous, adhesion at high deposition rates, and exceptional repeatability.

Limitations: relative complexity, expense, and parameter optimization.

Chemical method (sol-gel, colloidal): To produce metal particles, the colloidal approach reduces inorganic salts in solution chemically. Next, the arid substances consolidated, dried, and heated in a reducing atmosphere, like H2, to produce the metal component and encourage selective oxide reduction.

Using the sol-gel technique: two micelle solutions are made with 0.6 M NaBH4 (a), 0.1 M HAuCl4 (as), and mesoporous slice. and the mixture is mixed under UV light until the gold is completely reduced.

The iron's synthesis includes: For Fe/Au-containing nanocomposites, additional steps are required, such as the creation of a second shell, drying the powders following the second gold coating, pressing the mixture, and so forth.

Benefits include being straightforward, adaptable, having a low processing temperature, having strict stoichiometry control, and producing high-purity products.

Constraints: poor bonding, high permeability, limited wear resistance, and challenging porosity control.[17]

PMNCs or polymer matrix nanocomposites:

The use of pre-polymer/intercalation from solution is applicable to stratified reinforcing materials where intercalation of the polymer is possible. Primarily for layered silicates, where the prepolymer or polymer is intercalated from solution. Using a solvent that allows silicate layers to swell and the polymer or prepolymer to dissolve.

Advantages: low or even zero polarity polymers based on intercalated nanocomposites can be synthesized preparation of the filler in uniform dispersions.

Limitations: industrial solvent usage in huge quantities.

Inside-out intercalative polymerization: The layered silicate is encased in the monomer solution or liquid, which forms the polymer in the space between the incorporated sheets. Prior to the swelling process, polymerization can occur via radiation or heat, the diffusion of an appropriate initiator, or the application of a cation exchange-fixed catalyst within the interlayer. **Benefits**: The filler's dispersion in the polymers precursors makes the process simple. **Limitation**s: Intra-gallery polymerization is difficult to manage.

Melt intercalation: Annealing the polymeric material and the layered host in combination statically or under shear over the polymer's softening point. During the annealing process, polymer chains from the bulk polymer melt diffuse into the galleries between the host layers.

Benefits: include being safe for the environment, using polymers that are inappropriate for other processes, and being similar to industrial polymer methods. **Limitations**: Polyolefin, which makes up the majority of utilized polymers, has restricted applicability.

Template synthesis: The process of creating an inorganic material's layered structure in-situ. in a polymer-containing aqueous solution. The formation of strata is modeled by the water solubility. Extensively employed in the production of LDH nanocomposites, although layered silicates have seen less development. There are two varieties: in situ polymerization and mixing.

Benefits: straightforward process, large-scale production.

Limitations: few uses, mostly dependent on water-soluble polymers that are tainted by byproducts.

The By forming chemical bonds: The sol-gel technique embeds monomers and organic molecules on matrices to introduce organic groups forming the in-situ sol-gel matrix inside the polymer and/or simultaneously producing inorganic and organic networks.

Benefits include:

Being easy to use, adaptable, having a high degree of producing high-purity products, maintaining stringent stoichiometry control, creating chemical homogeneity, and creating. Threedimensional polymers with connections between metal and oxygen. One matrix or several. Especially helpful when utilizing liquids or viscous fluids to create composite materials.

Limitations: Compared to the mixing process, there is more shrinkage and less voids.

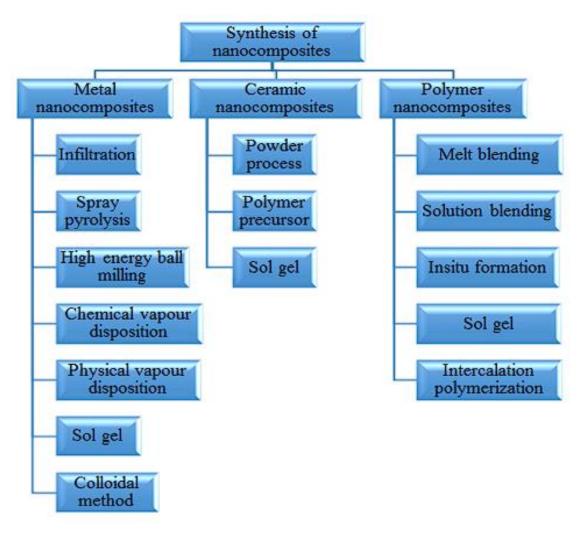


Fig 2.Synthesis of nanocomposites

APPLICATION:

With their extraordinary design potential and hitherto unheard-of property combinations, polymer nanocomposites are becoming the high-performance materials of the twenty-first century. Utilized in a wide range of cutting-edge technologies. Polymer types utilized in nanocomposite.[18]

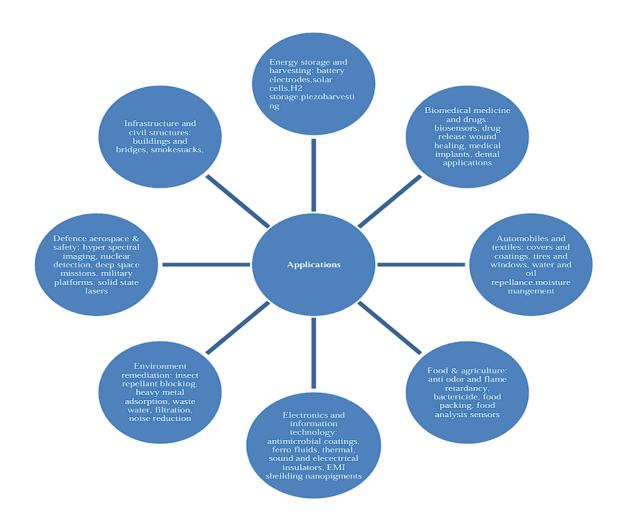


Fig. 3. Application of Nanocomposite

Classification of vinyl polymers: Methyl methacrylate (MMA), MMA copolymers, polyacrylonitrile, and polyaacrylic acid are examples of addition polymers. Other examples are Poly (ethylene glycol), Poly(ethylene vinyl alcohol), Poly(vinylidene fluoride), Poly(p-phenylenevinylene), Polybenzoxazole, and Poly(S-co-AN). [19]

2. Classification of rubbers and condensation polymers: Other polyamides, nylon-6, poly(E-caprolactone) (PCL), poly(ethylene terephthalate) (PET), poly(trim ethylene terephthalate), poly(butylenes terephthalate), polycarbonate, polyethylene oxide, polyethylene oxide copolymers, poly(ethylene immune), poly (dim ethyl siloxane), polybutadiene, copolymers of polybutadiene, deoxidized natural rubber, epoxy polymer resins, phenolic resins, polyurethanes, polyurethane urea, polyimide's, poly(amice acid), polysulphone, polyetherimide, fluoropoly(ether-imides).[20]

Polyolefin classification: Poly (ethylene-propylene diene ethylene linkage), Poly (ethylene-co-vinyl acetate), Poly (PP), PE, and PE oligomer (1-butene) 1. Specialty polymers: liquid crystalline polymers, hyper branch polymers, acryl-ethany-terminated imide oligomer, cyanate ester, Nafion oly (p-phenylene vinylene), poly (N-vinylcarbazole), polyaniline, a substance called poly and hyper branch polymers. Polylactide (PLA), Poly (butylene carboxylate), PCL, saturated polyester, Polyhydroxy butyrate and aliphatic polyester are examples of biodegradable polymers. [21]

The future of nanocomposites as: These materials have an increasing number of real-world applications. According to reports, the global output is expected to surpass 600,000 tons in less than two years and will encompass the ensuing five to ten years in the following areas. [22]

- I. Methods of drug delivery.
- II. UV protection gels.
- III. Anti-corrosion barrier coatings.
- IV. Lubricants and scratch-free paints.
- V. Novel fire-retardant materials
- VI. New materials that withstand abrasion and scratches.
- VII. Fibers and films with superior strength.

Nanocomposites materials are of great interest in many automotive and general/industrial applications due to their improved mechanical properties. These include the capacity to be utilized on various vehicle types like door handles, timing belt covers, intake manifolds, engine covers, and mirror housings. The oxygen, carbon dioxide, moisture, and odor barrier qualities of nanocomposites are significantly enhanced. They also retain impact strength and film clarity while exhibiting higher stiffness, strength, and heat resistance. These novel materials and their industrial uses are becoming more apparent to the manufacturing sector. Materials science is undergoing a revolution thanks to nanotechnology. Despite being important applications in many industrial domains, nanocomposites are not yet widely commercialized due to a number of significant technological and financial obstacles. Future applications of biodegradable polymer-based nanocomposites as advanced biodegradable materials exhibit great potential. These materials, which come from botanical and natural sources, are whole new categories Upcoming.

Developments call for the application of this nanotechnology to other kinds of polymer systems, which will probably require the creation of fresh compatibility techniques. When working with nanomaterials, there are some safety factors to take into account. For example, the release of nanoparticles into the atmosphere poses a serious risk to public health and safety. Studies on the emission of nanocomposites and their effects on the environment are therefore crucial. This could be because of the potentially harmful properties of products derived from nanotechnology, such as their crystalline form, reactivity, and huge surface area, which may facilitate their transit into the environment or let them to interact with components of cells, thereby exacerbating a number of detrimental consequences associated with their composition.[23]

FINAL SUMMARY

Difficulties in Nanocomposite Processing and Manufacturing:

The field of conventional composite materials could be completely redefined by nanocomposite materials in terms of performance and future use. One of the biggest obstacles, nevertheless, will be to develop those processing-manufacturing technologies in terms of quantity, quality, and value for commercialization.[24]

Among those crucial concerns are still the dispersion of nanoparticles and their physical compatibility with matrix materials. Before realizing the entire potential of CNF nanocomposites, the goal of addressing those interfacial adhesion problems and total dispersion in the carbon nanofibers matrix needs to be addressed.[25]

The definition, history, categorization, synthesis, characteristics, polymers employed in the composition, applications, and potential future uses of nanocomposites were all covered in this study. The "nano effect" causes materials to exhibit some peculiar and exotic properties when they are shrunk to the nanoscale. The development of nanocomposites enables the creation of components with distinct mechanical and physical attributes. It is anticipated that nano macroscopic composite-based sensors will have a substantial effect on food security, security surveillance, environmental monitoring, and clinical diagnosis. Substances such as nanocomposites can be used to satisfy new needs brought about by technological and scientific advancements. Therefore, all three forms of nanocomposites offer benefits and potential, piquing interest in these novel materials around the globe. [26]

Future Perspectives: The cannot overstate the importance of biorenewable materials in achieving the SDGs, particularly in light of the present state of mental health issues, energy security, and local economic growth linked to the creation of jobs and the preservation of foreign exchange. However, there are a number of barriers that prevent biorenewable from being used in both home and commercial settings. The quick biodegradation of bore holes in target habitats is one such difficulty, but it can be overcome by altering the materials' surface and chemical characteristics to lengthen their biopersistence. However, this disadvantage can be overcome by gaining a fundamental understanding of their biosynthesis and inherent chemical properties that could be adjusted to improve the physiochemical and mechanical characteristics of these materials. Another issue is the inferior physiochemical and mechanical properties of biorenewable compared with those of the traditional materials obtained from fossil hydrocarbons. Furthermore, in order to ensure the best possible interactions between the matrix and fillers as well as material dissolution and regeneration/reconstitution, it is necessary to properly understand fundamental theories, new insights, and more sophisticated processing technologies. These factors include the surface properties and solubility of biorenewable materials as well as their interactions with solvents, additives, and/or fillers. In order to

determine the kinetic parameters and reaction mechanisms of biorenewable nanomaterials in diverse settings, improved characterisation approaches are necessary. Furthermore, despite its importance for improving the mechanical properties of biomasses nanocomposite materials, the theory of chain dynamics and mobility is still speculative and not fully understood.

References :

- 1. Ajayan PM, Schalder LS, Braun PV. 2003.
- 2. Rozenberg BA, Tenne R. 2008; 33(1): p. 40-112.
- Pedro Henrique Cury Camargo, Kestur Gundappa Satyanarayana, Fernando Wypych, 2009, "Nanocomposites: Synthesis, Structure, Properties and New Application Opportunities (Review Articles), Materials Research, Vol. 12, No. 1, 1-39.
- 4. Rozenberg BA, Tenne R. 2008; 33(1): p. 40-112.
- Kamigaito O, 1991, "What can be improved by nanometer composites?" Journal of Japan Society of Powder Metallurgy, 38:315-321.
- Schmidt D, Shah D, Giannelis EP, 2002, "New advances in polymer/layered silicate nanocomposites", Current Opinion in Solid State & Materials Science; 6(3):205-212.
- Pandey JK, Kumar AP, Misra M, Mohanty AK, Drzal LT, Singh RP, 2005, "Recent advances in biodegradable nanocomposites", Journal of Nanoscience and Nanotechnology; 5(4):497-526.
- Farzana Hussain, Mehdi Hojjati, Masami, Okamoto, Russell E. Gorga, 2006, "Review Article: Polymer-matrix Nanocomposites, Processing, Manufacturing, and Application: An Overview", Journal of Composite Materials, Vol. 40, No. 17, DOI: 10.1177/0021998306067321
- 9. Oriakhi, C.O, 1998, "Nano Sandwiches", Chem. Br., 34: 59-62.
- 10. Pedro H, Cury CK, Gundappa S, Fernando W. 2009; 12(1): p. 1-39.
- 11. Kamigaito O, 1991, "What can be improved by nanometer composites?" Journal of Japan Society of Powder Metallurgy, 38:315- 321.
- 12. Iijima S, 1991, "Helical microtubes of graphitic carbon", Nature, 354(6348):56-58.
- 13. Biercuk MJ, Llaguno MC, Radosvljevicm, HJ, 2002, "Carbon nanotube composites for thermal management", Appied Physics Letters; 80(15):2767-2769.
- Ounaies Z, Park C, Wise KE, Siochi EJ, Harrison JS, 2003, "Electrical properties of single wall carbon nanotube reinforced polyimide composites", Composites Science and Technology; 63(11):1637-1646.

- Weisenberger MC, Grulke EA, Jacques D, Ramtell T, Andrews R, 2003, "Enhanced mechanical properties of polyacrylonitrile: multiwall carbon nanotube composite fibers", Journal of Nanoscience and Nanotechnology; 3(6):535-539.
- 16. Schmidt D, Shah D, Giannelis EP, 2002, "New advances in polymer/layered silicate nanocomposites", Current Opinion in Solid State & Materials Science; 6(3):205-212.
- Wypych F, Seefeld N, Denicolo I, 1997, "Preparation of nanocomposites based on the encapsulation of conducting polymers into 2H-MoS2 and 1T-TiS2", Quimica Nova; 20(4):356-360.
- Aruna ST, Rajam KS, 2003, "Synthesis, characterisation and properties of Ni/PSZ and Ni/YSZ nanocomposites", Scripta Materialia; 48(5):507-512.
- Giannelis EP, 1996, "Polymer layered silicate nanocomposites", Advanced Materials; 8(1):29-35.
- Wypych F, Adad LB. Grothe MC, 1998, "Synthesis and characterization of the nanocomposites K-0, K0, 1(PEO) xMoS2 (X = 0,5; 1,2)", Quimica Nova; 21(6):687-692. [12].
- Sternitzke M, 1997, "Review: structural ceramic nanocomposites", Journal of European Ceramic Society; 17(9):1061-1082.
- 22. Peigney A, Laurent CH, Flahaut E, Rousset A, 2000, "Carbon nanotubes in novel ceramic matrix nanocomposites", Ceramic International; 26(6):677-683.
- Alexandre M, Dubois P, 2000, "Polymer-layered silicate nanocomposites: preparation, properties and uses of a new class of materials", Materials Science & Engineering; 28(1-2):1-63.
- 24. Gangopadhyay R, Amitabha D, 2000, "Conducting polymer nanocomposites: a brief overview", Chemistry of Materials; 12(7): 608-622.
- Thostenson ET, Ren Z, Chou TW, 2001, "Advances in the science and technology of carbon nanotubes and their composites: a review", Composites Science & Technology; 61(13):1899-1912.
- Pokropivnyi VV, 2002a, "Two-dimensional nanocomposites: photonic crystals and nanomembranes (review): Types and preparation", Powder Metallurgy and Metal Ceramics; 41(5-6):264-272.
- Pokropivnyi VV, 2002b, "Two-dimensional nanocomposites: photonic crystals and nanomembranes (review). II: Properties and applications", Powder Metallurgy and Metal Ceramics; 41(7-8):369-381.
- 28. Pedro H, Cury CK, Gundappa S, Fernando W. 2009; 12(1): p. 1-39.
- 29. Giannelis EP. 1996; 8(1): p.29-35.

- 30. Theng BKG. 1974.
- 31. Livage J. 1997; 2(2): 132-136
- 32. Hench LL, West JK.1990; 90(1): 33-72.
- 33. Mathur S, Veith M, Shen H, Huner S, Jilavi M. 2002; 14(2): p. 568-582.
- 34. Erik T Thostenson, Chunyu Li, Tsu-wei Chou. 2005; 65: p.491-516.
- 35. Paul DR, Robeson LM. 2008; 49: p. 3187-3204